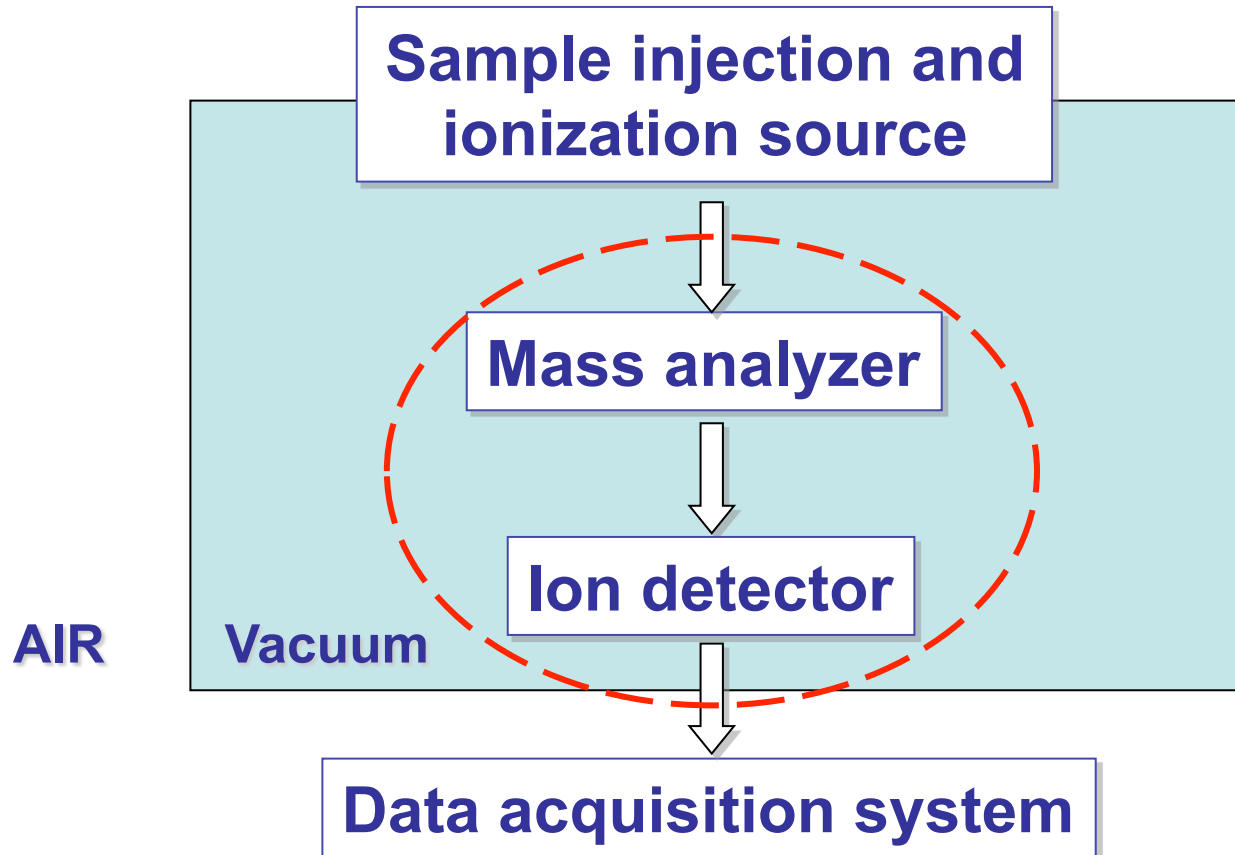
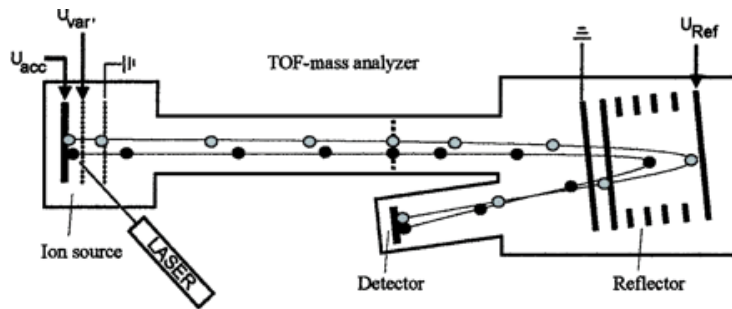


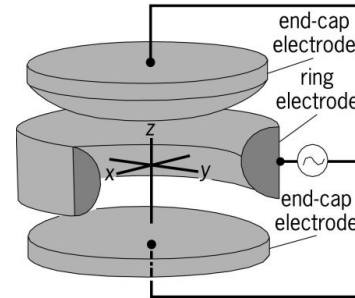
General Configuration of Mass Spectrometer



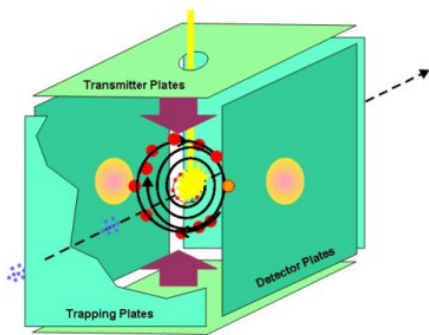
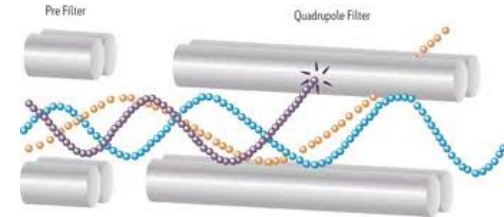
Important Mass Analyzers



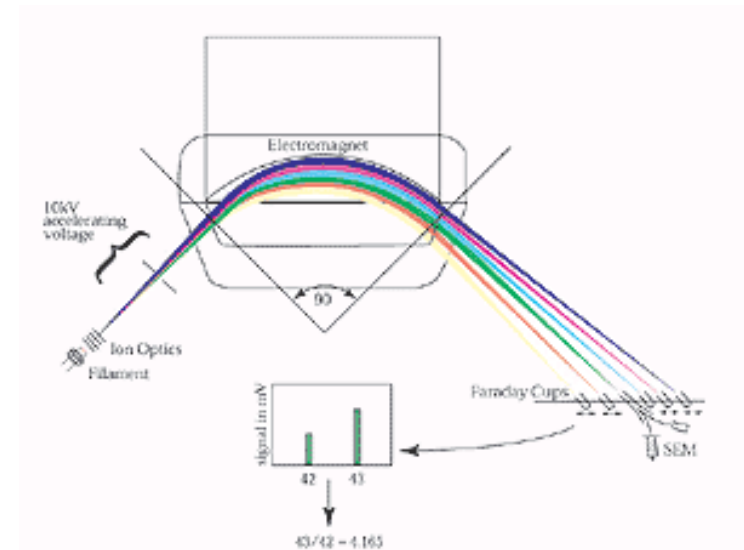
Time-of-Flight (TOF)



Ion Trap/Quadrupole

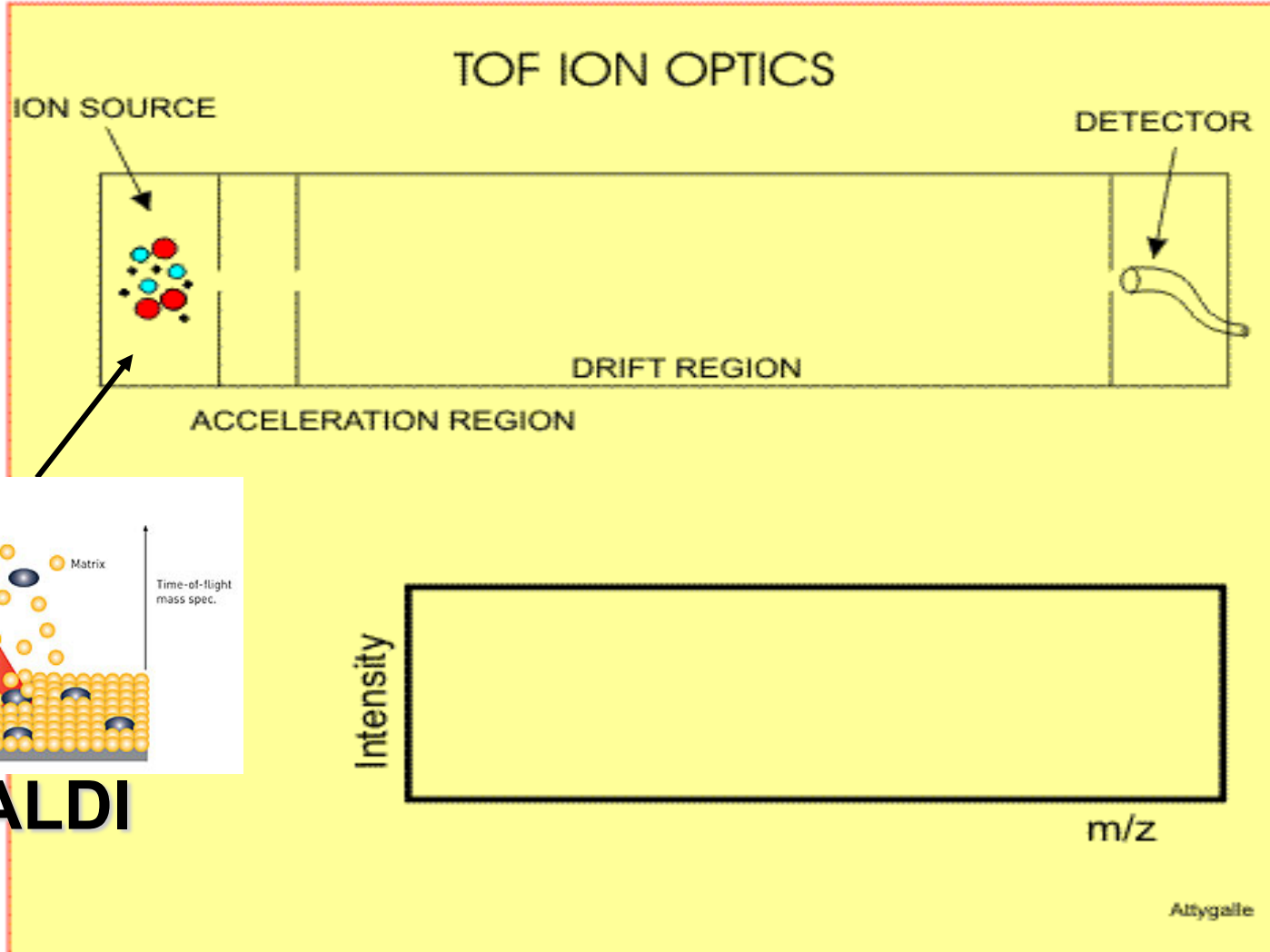


Fourier-Transform Mass Analyzer



Magnet Sector

Time-Of-Flight Mass Analyzer (TOF)



MALDI-TOFMS

Principle

Energy conservation between potential and kinetic energies

Kinetic energy (K.E.) = qV

$$K.E. = (1/2) m v^2 = qV$$

$$v = (2qV/m)^{1/2}$$

The traveling time (t) through the drift region is L/v where L is the length of the drift region.

$$t = L / (2V/m/q)^{1/2}$$

Parameters:

q = electric charge on ions

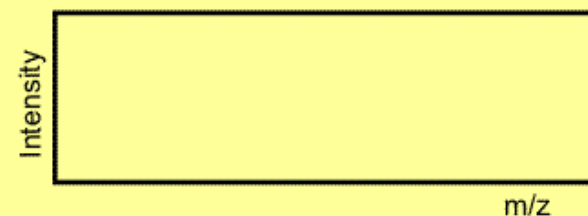
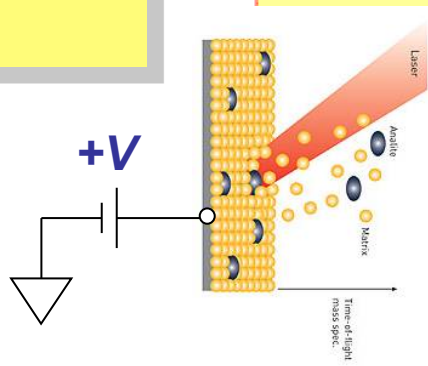
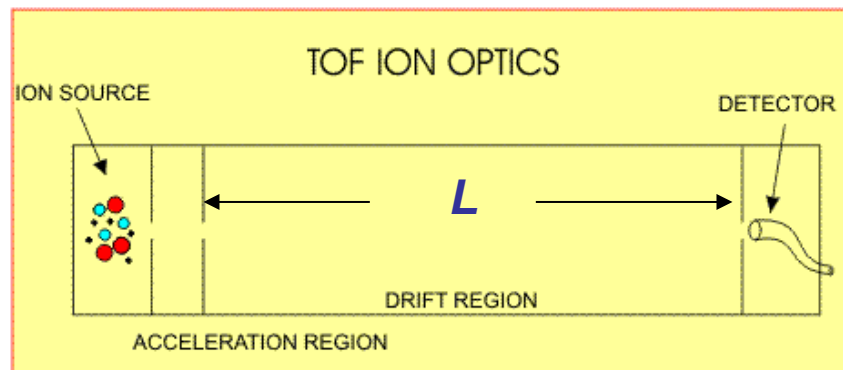
V = Acceleration voltage for ions

m = molecular weight of ions

v = velocity of ions

t = traveling time

L = drift length

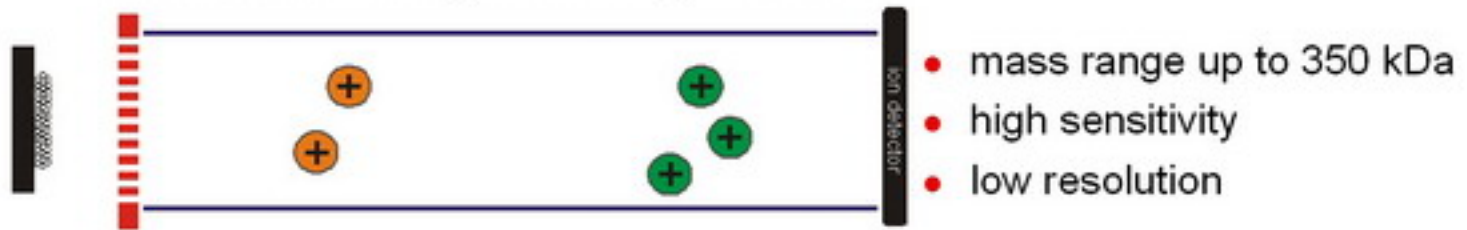


Attygalle

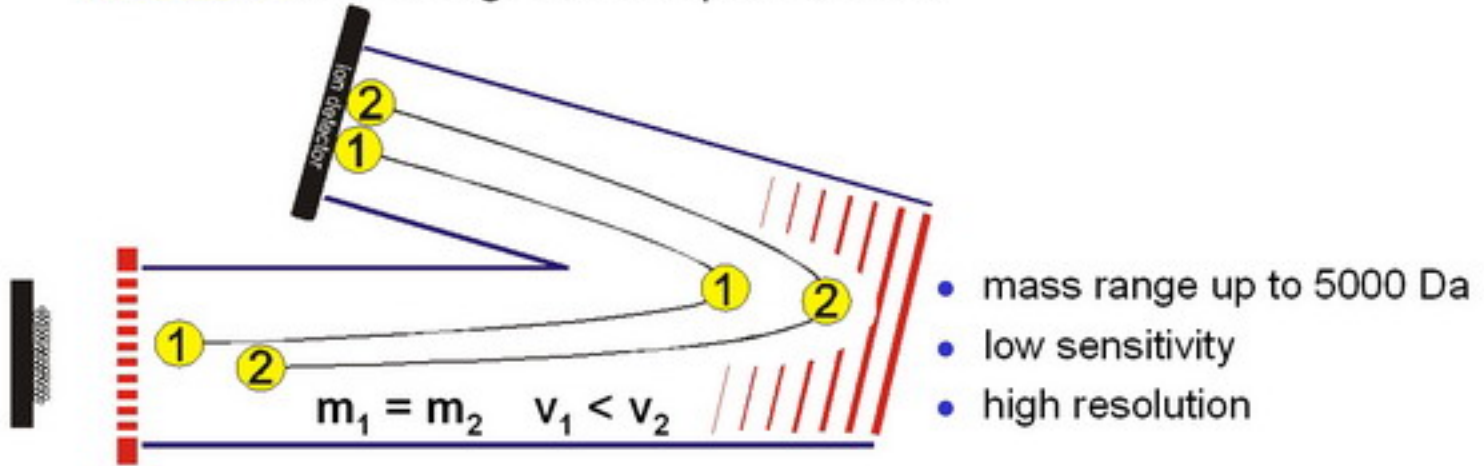
Commercial TOF

Linear and reflector TOF MS

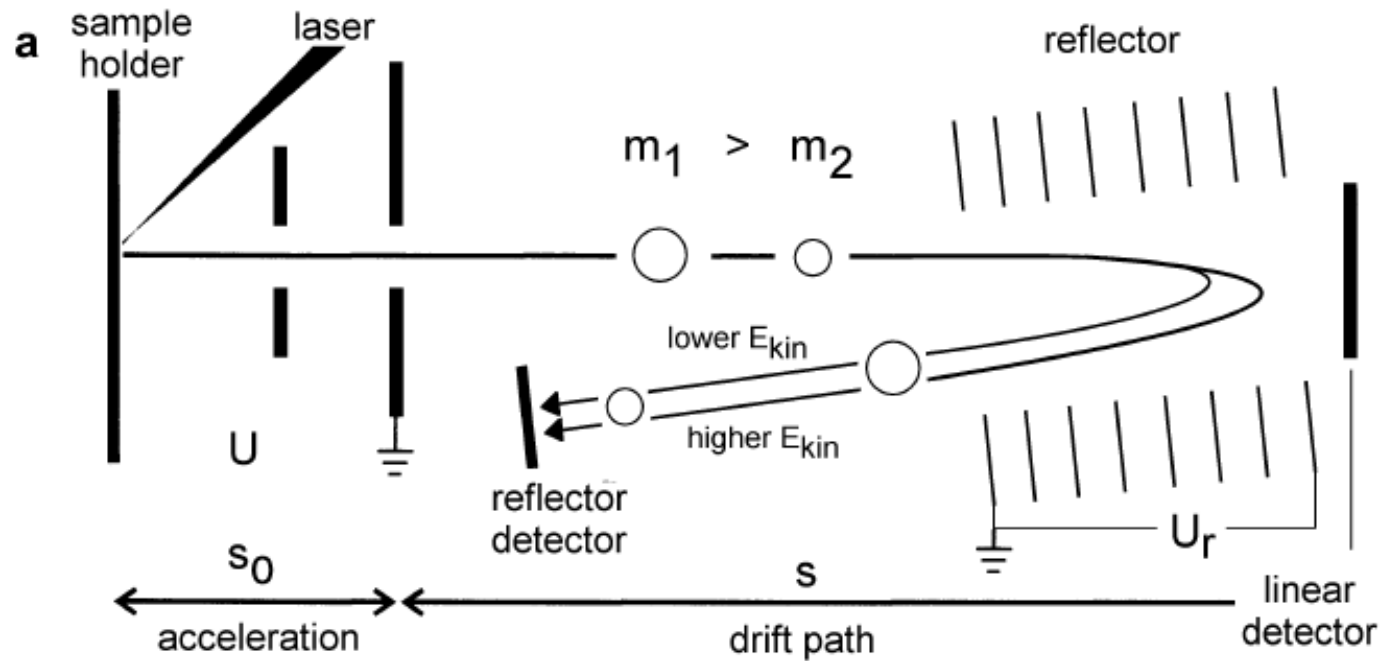
Linear time-of-flight mass spectrometer

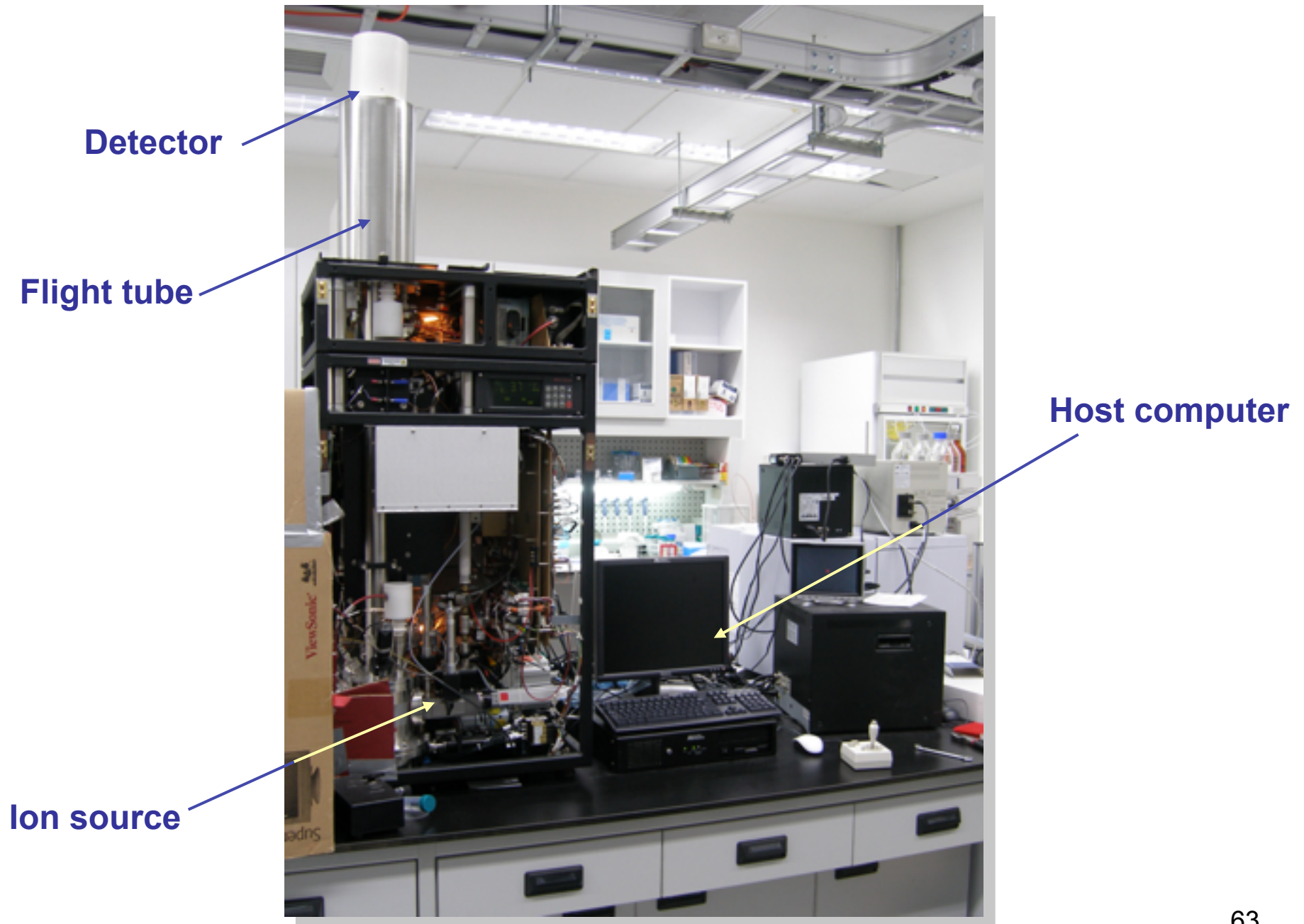


Reflector time-of-flight mass spectrometer



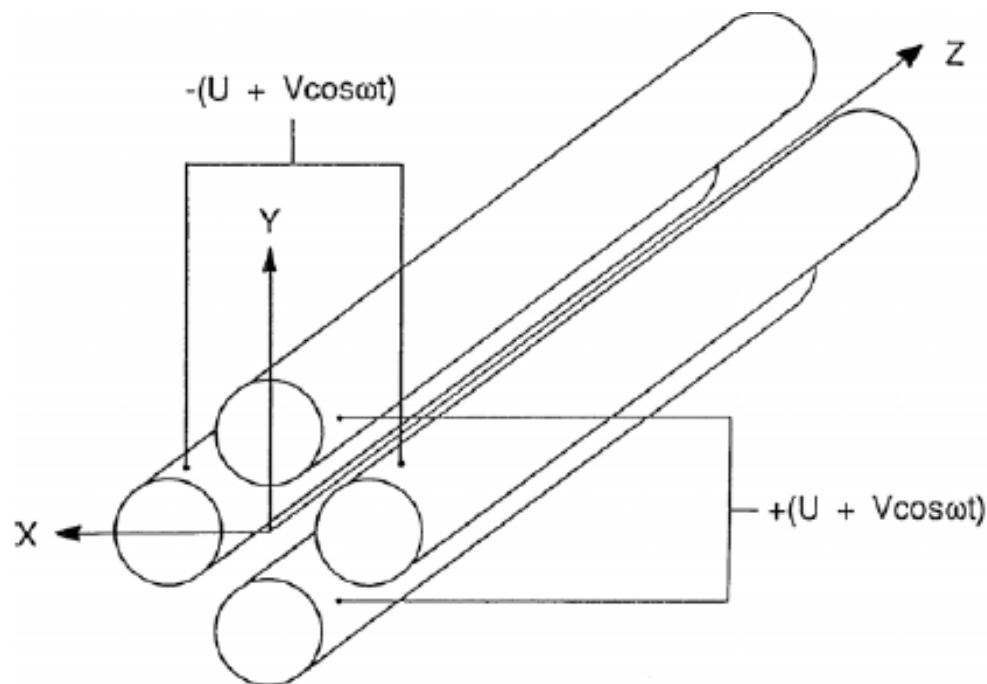
Commercial TOF





MALDI-TOF MS (*Apply Biosystem Inc., USA*)

Quadrupole Mass Filter



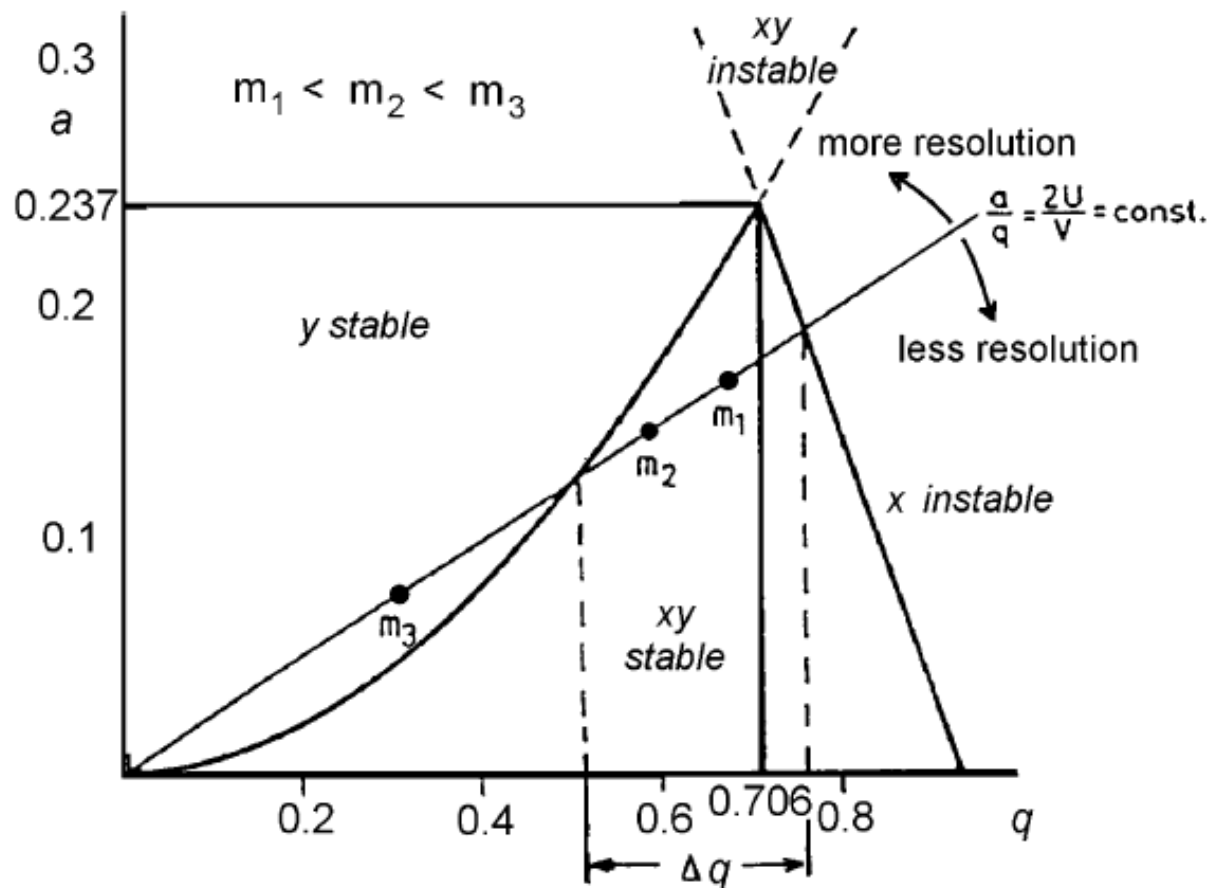
Equation of ion motion (Mathieu eq.)

$$\frac{d^2 x}{dt^2} + \frac{e}{m_i r_0^2} (U + V \cos \omega t) x = 0$$

$$\frac{d^2 y}{dt^2} - \frac{e}{m_i r_0^2} (U + V \cos \omega t) y = 0$$

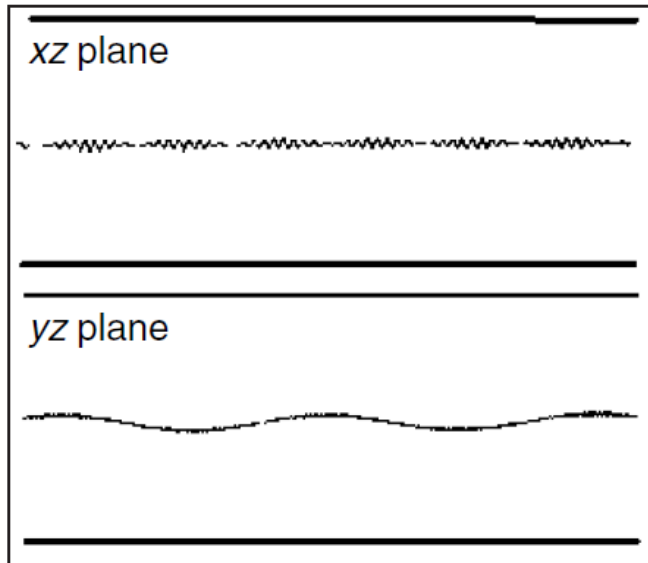
$$a_x = -a_y = \frac{4eU}{m_i r_0^2 \omega^2}, \quad q_x = -q_y = \frac{2eV}{m_i r_0^2 \omega^2}$$

Quadrupole Mass Filter



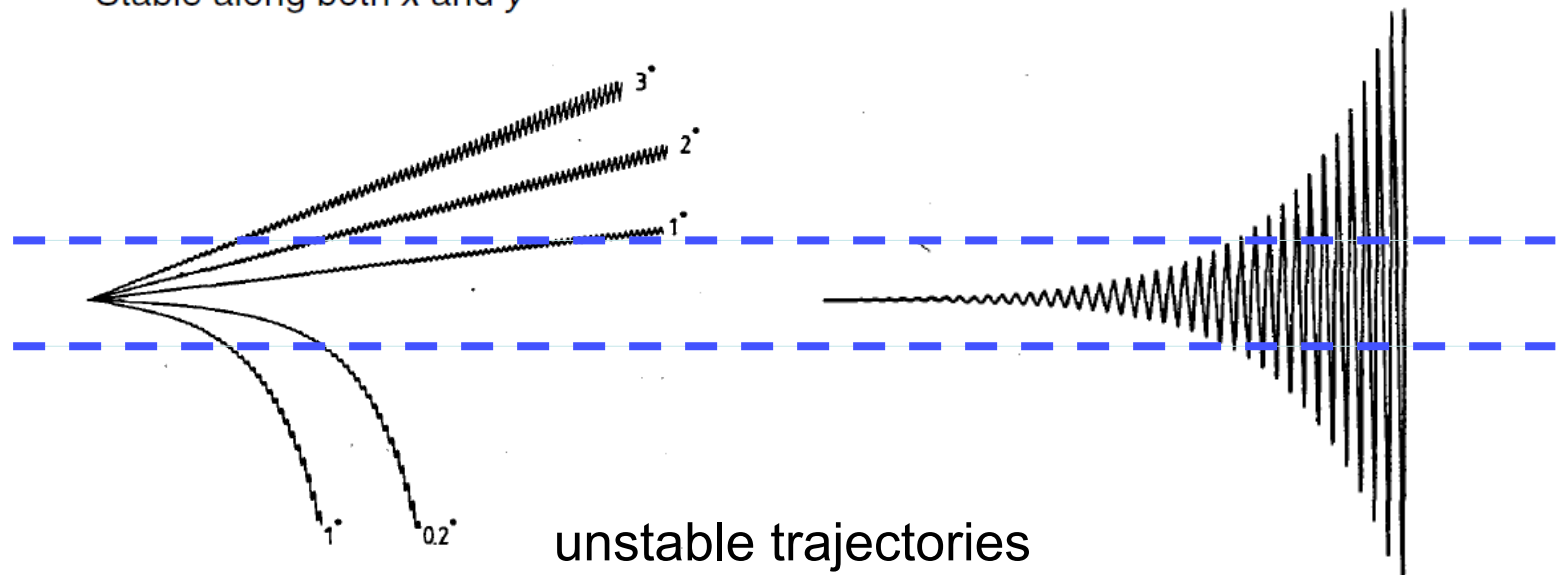
$$a_x = -a_y = \frac{4eU}{m_i r_0^2 \omega^2}, \quad q_x = -q_y = \frac{2eV}{m_i r_0^2 \omega^2}$$

Quadrupole Mass Filter

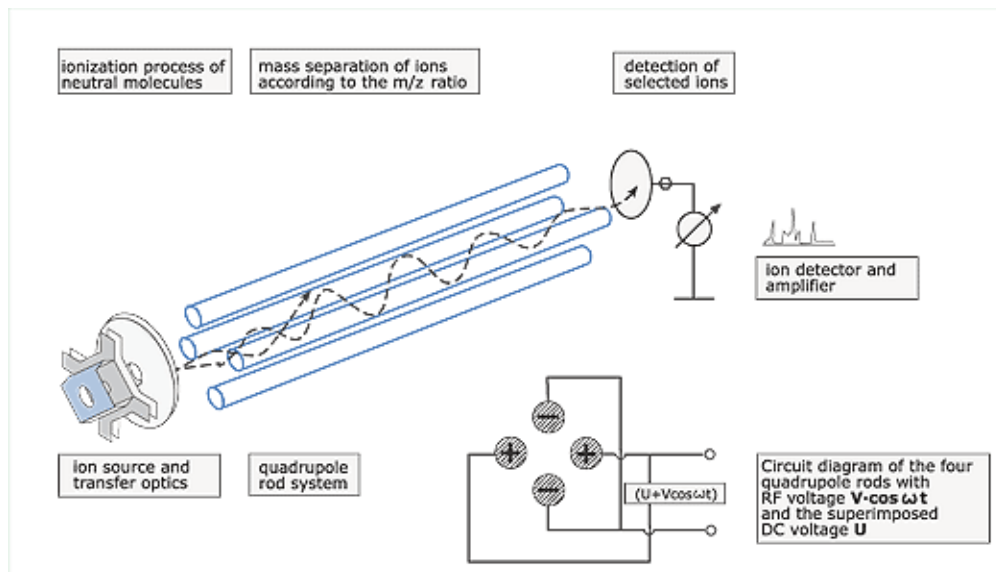


Stable along both x and y

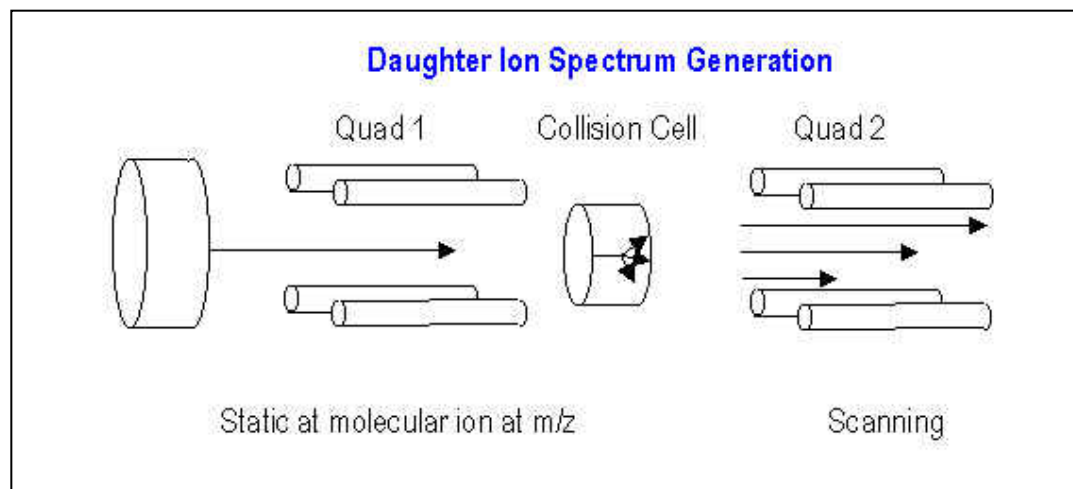
stable trajectory



Quadrupole Rods and Triple-Quad MS



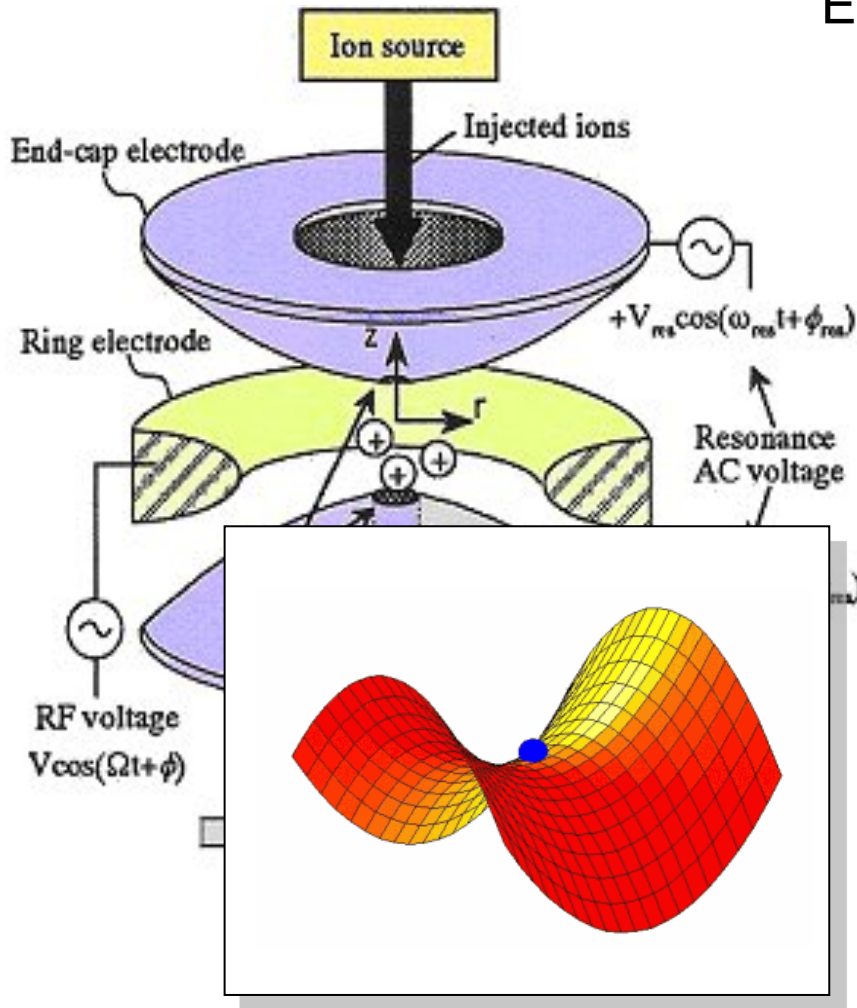
Inprocess Instruments



Dr. Graeme T. Clark , University Child Health, Southampton General Hospital,

Quadrupole Ion Trap

Equation of ion motion (Mathieu eq.)



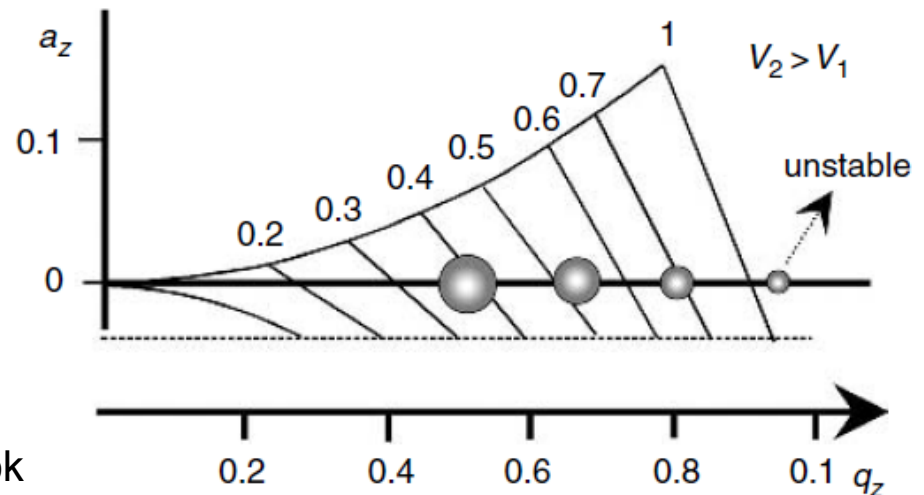
$$q_z \leq 0.908$$

$$\frac{d^2 z}{dt^2} - \frac{4e}{m_i (r_0^2 + 2z_0^2)} (U - V \cos \Omega t) z = 0$$

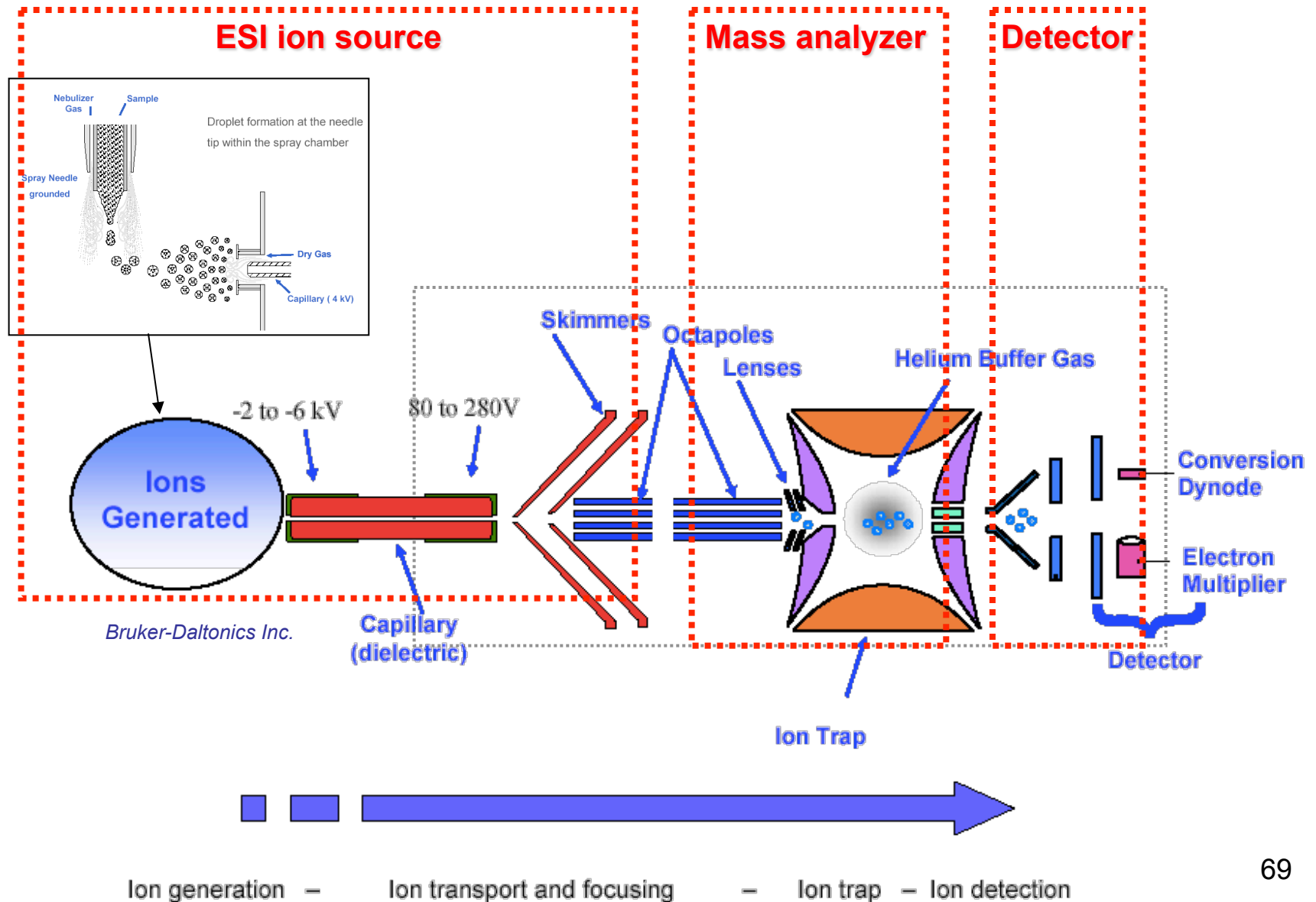
$$\frac{d^2 r}{dt^2} + \frac{2e}{m_i (r_0^2 + 2z_0^2)} (U - V \cos \Omega t) r = 0$$

$$a_z = -2a_r = -\frac{16eU}{m_i (r_0^2 + 2z_0^2) \Omega^2}$$

$$q_z = -2q_r = \frac{8eV}{m_i (r_0^2 + 2z_0^2) \Omega^2}$$



Ion Trap (with ESI source)



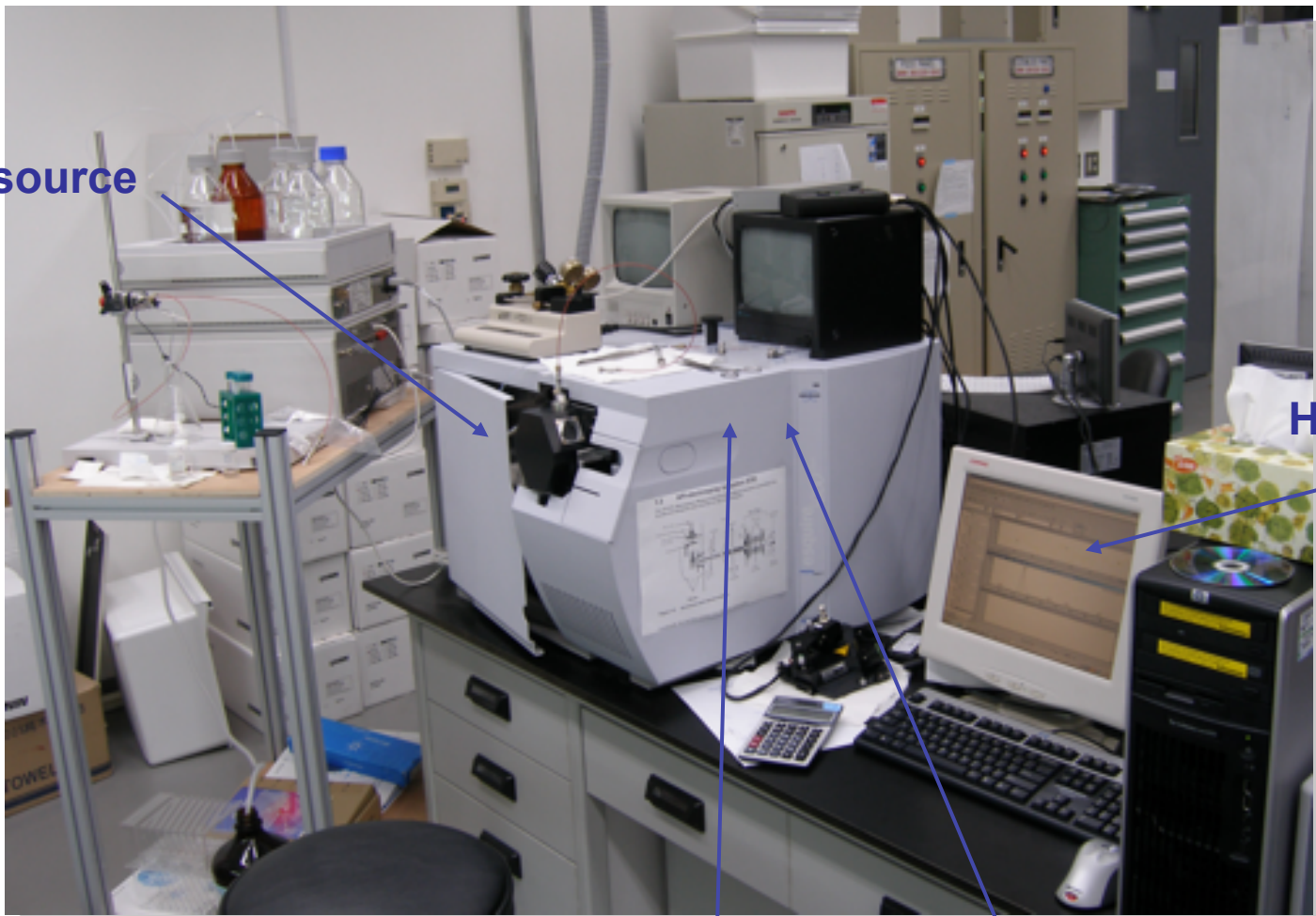
ESI-Q MS (*Bruker-Daltonics Inc., USA*)

Ion source

Host computer

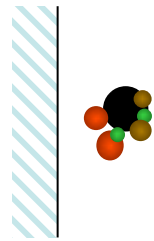
Ion trap

Detector



Sequential V.S. Simultaneous Detection

Sequential Detection

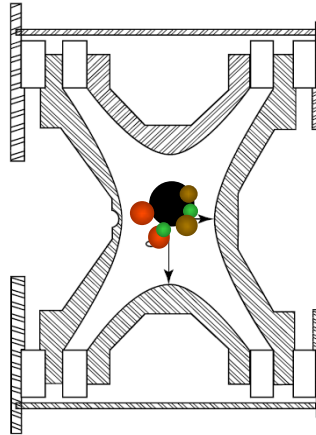


TOF MS



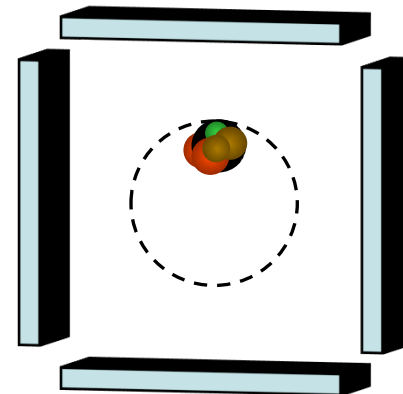
Detector

QIT MS



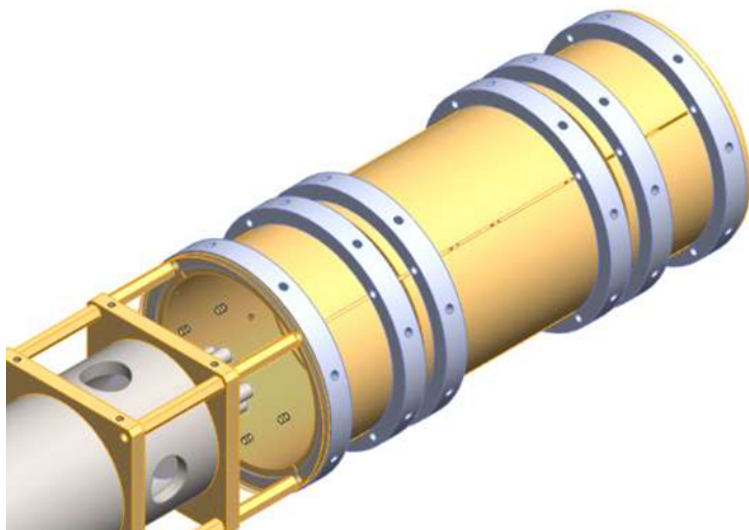
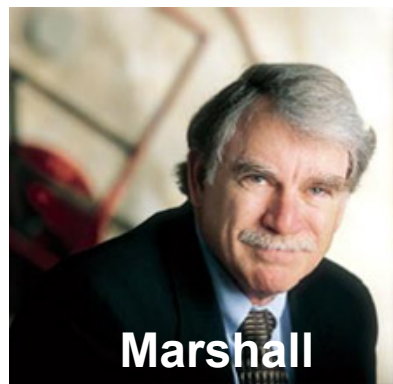
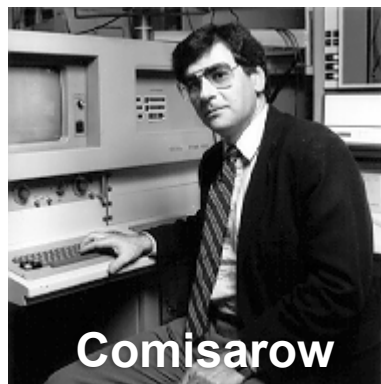
Detector

Simultaneous Detection
(Fourier-Transform MS)

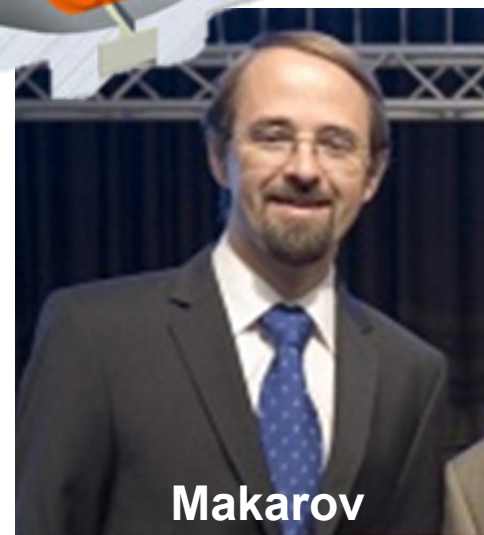
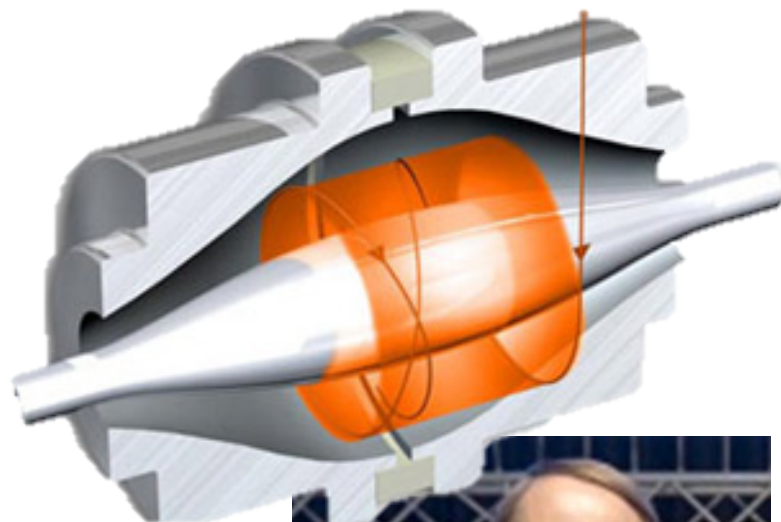


Commercial FT Mass Spectrometers

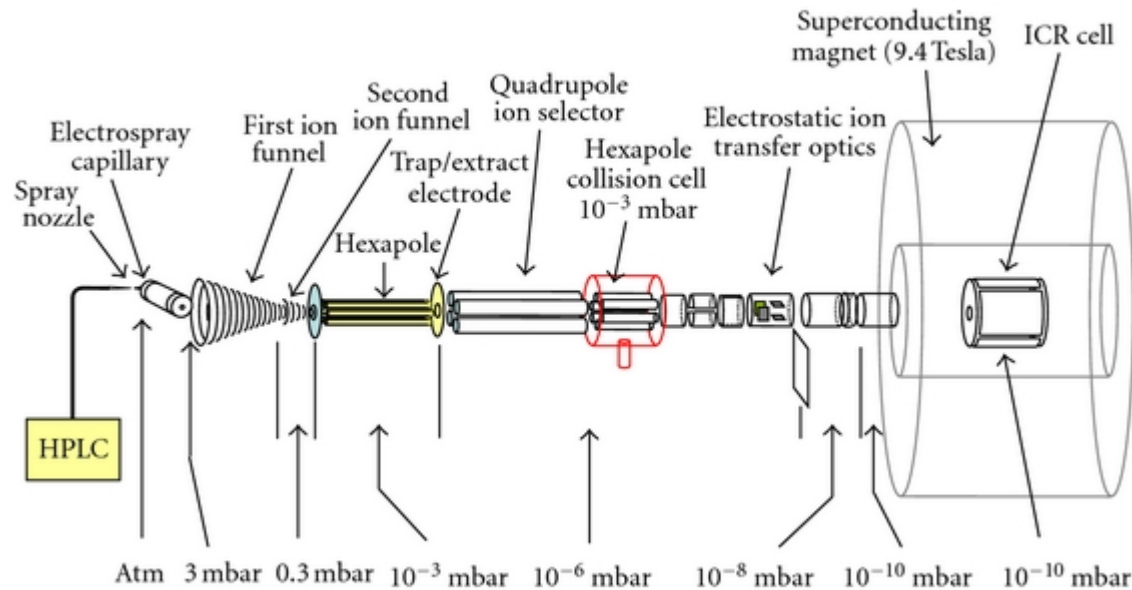
FT-ICR



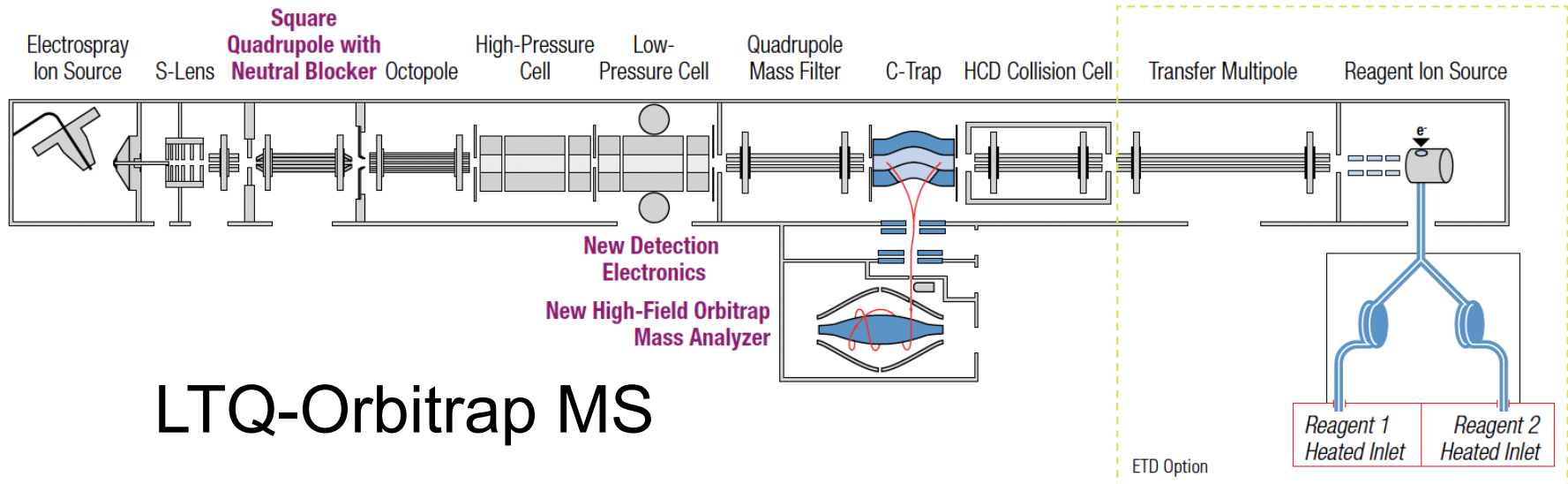
Orbitrap



Configuration of FTMS



FT-ICR MS



LTQ-Orbitrap MS

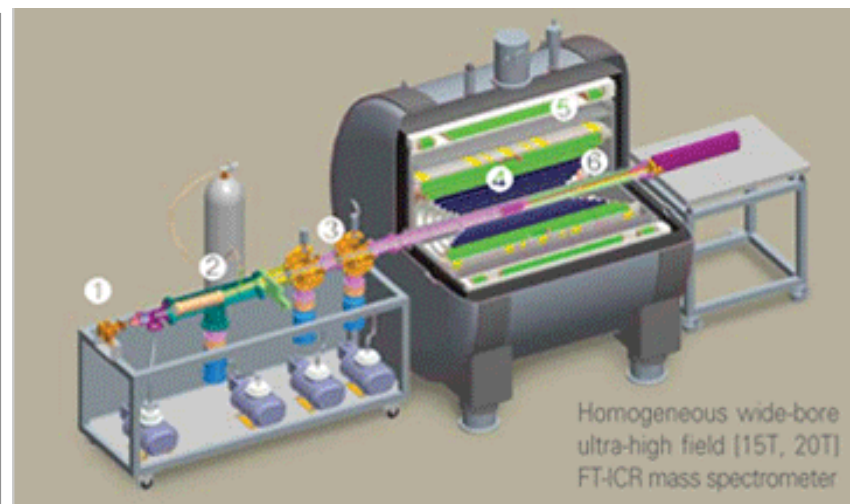
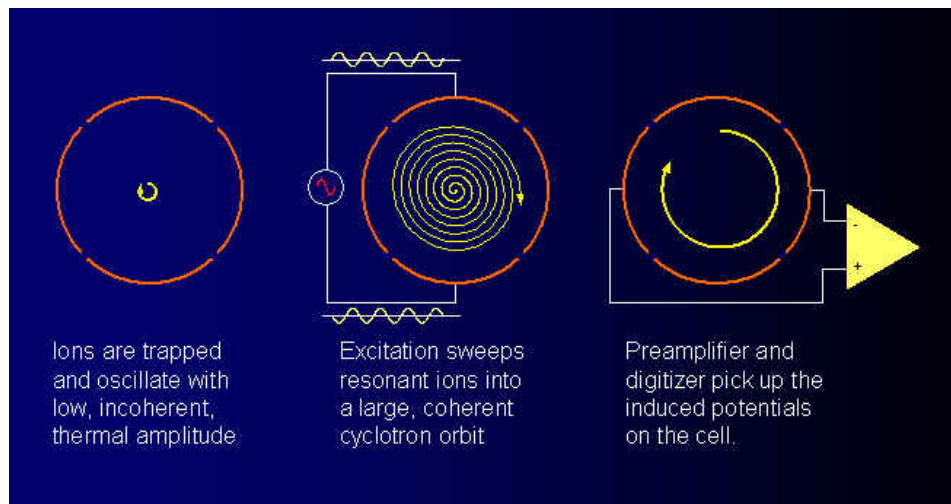
FT-ICRMS

Fourier-Transform

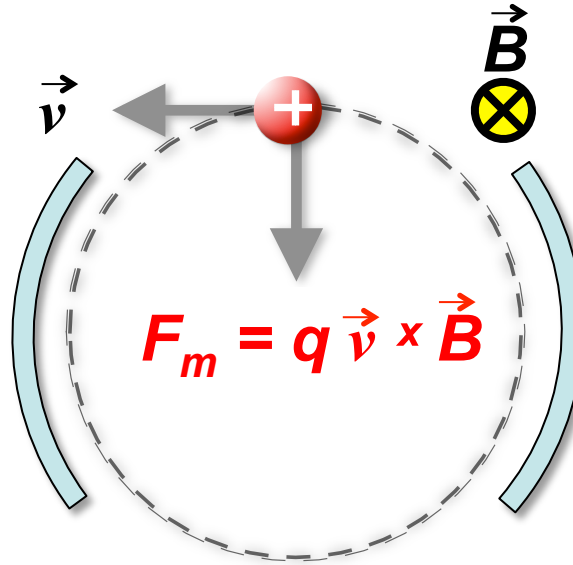
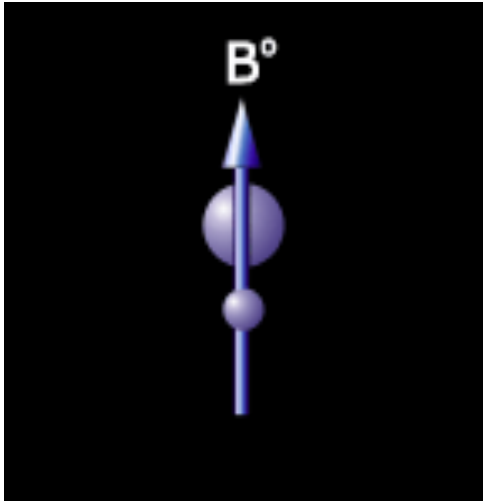
A method for data handling, which converts time-domain signal to frequency-domain data using Fourier transformation method.

Ion-Cyclotron-Resonance

High resolution ion recognition by detecting the frequency of ion cyclotron motion in high magnetic field.



Ion-Cyclotron Motion in High Magnetic Field



$$F_m = m \vec{v}^2 / r = q \vec{v} \vec{B}$$

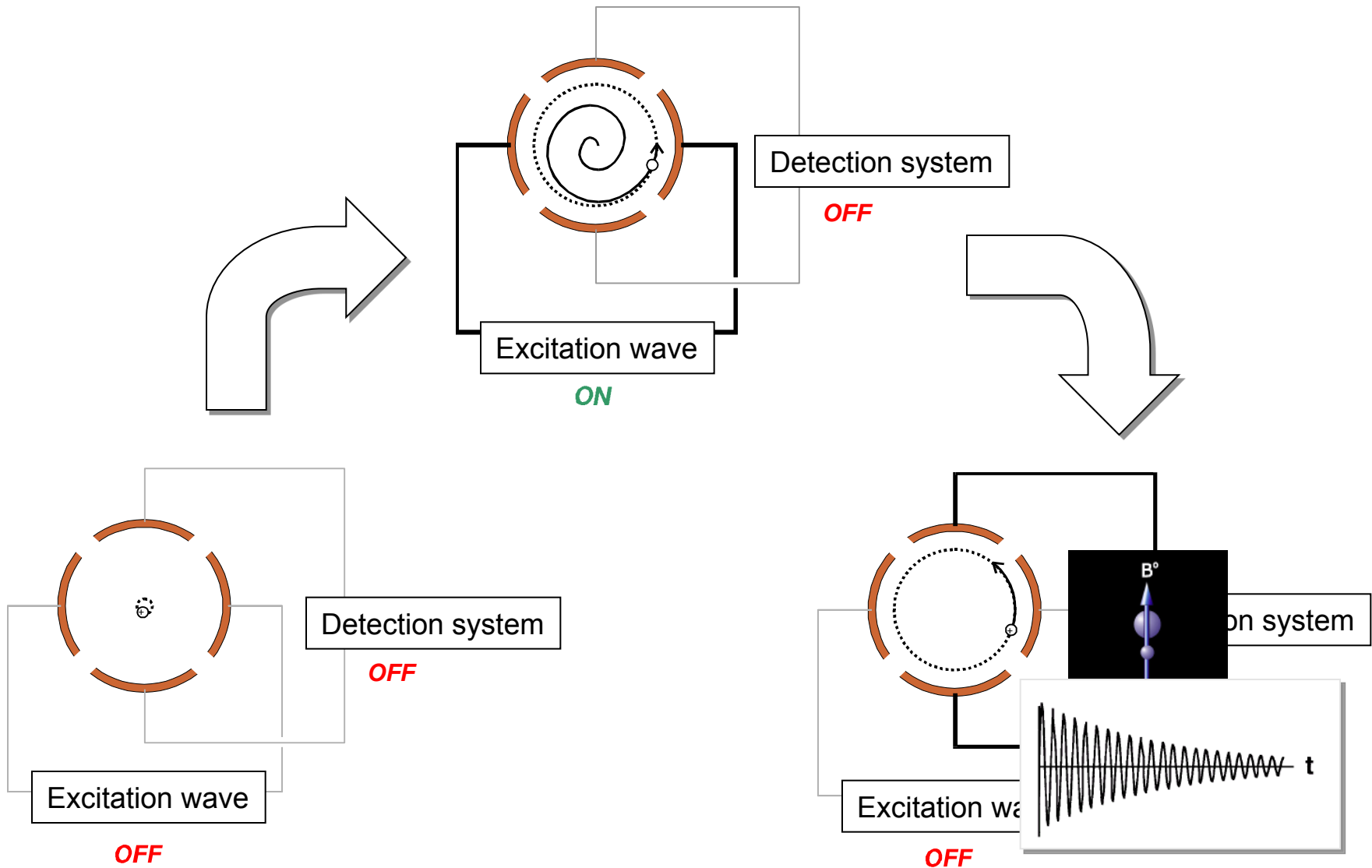
$$m \vec{v} / r = m \omega_c = q \vec{B}$$

$$\omega_c = (q/m) \vec{B}$$

$$\nu_c = \omega_c / 2\pi$$

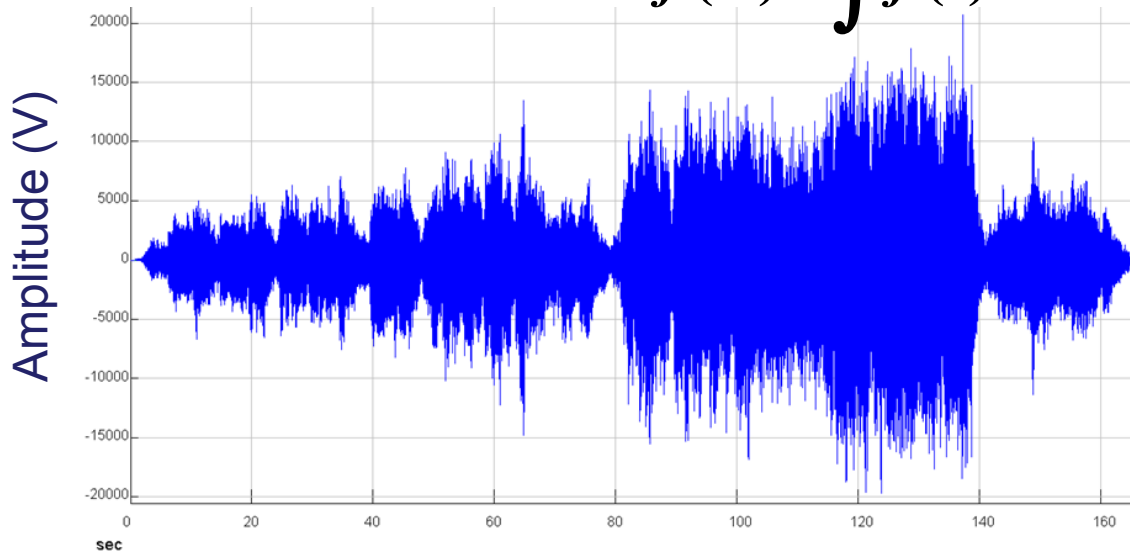
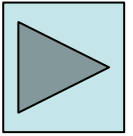
An ion moving in the presence of a spatially uniform magnetic field B is subject to a force (F).

Ion Cyclotron Motion: Excitation and Detection



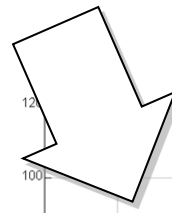
The Concept of Fourier-Transform in MS

$$f(\omega) = \int f(t)e^{-i\omega t} dt$$



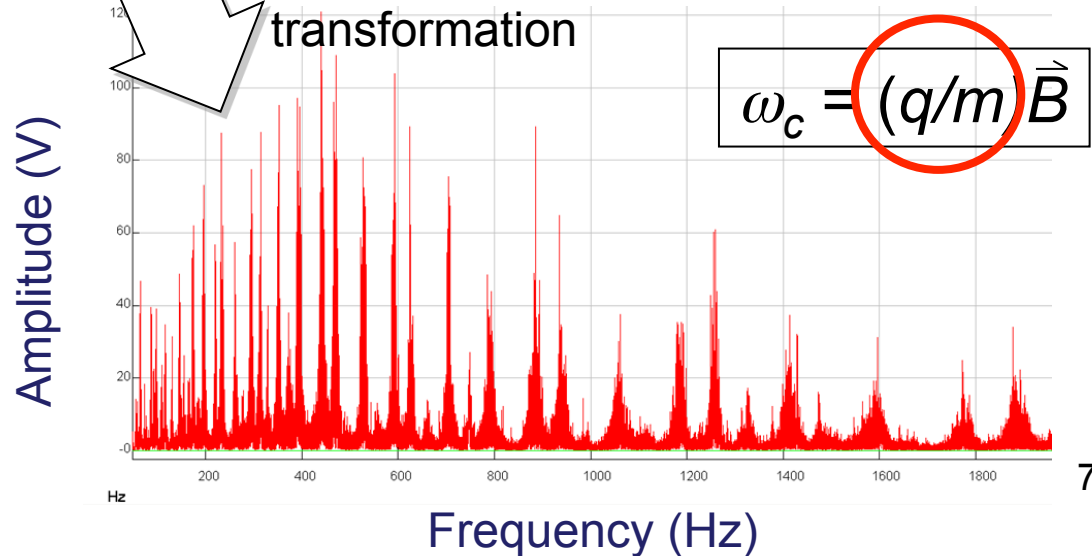
signal in
time-domain
(時域)

Time (s)



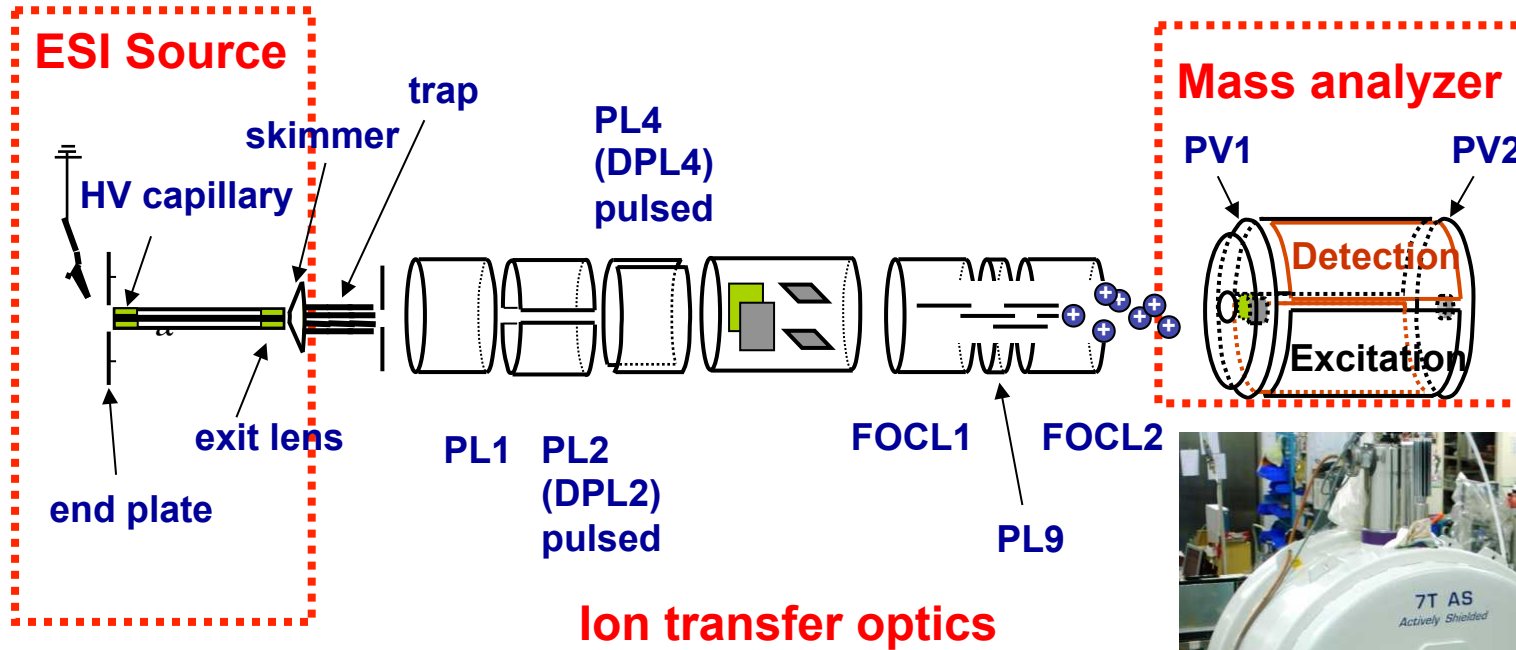
Fourier
transformation

signal in
frequency-domain
(頻域)



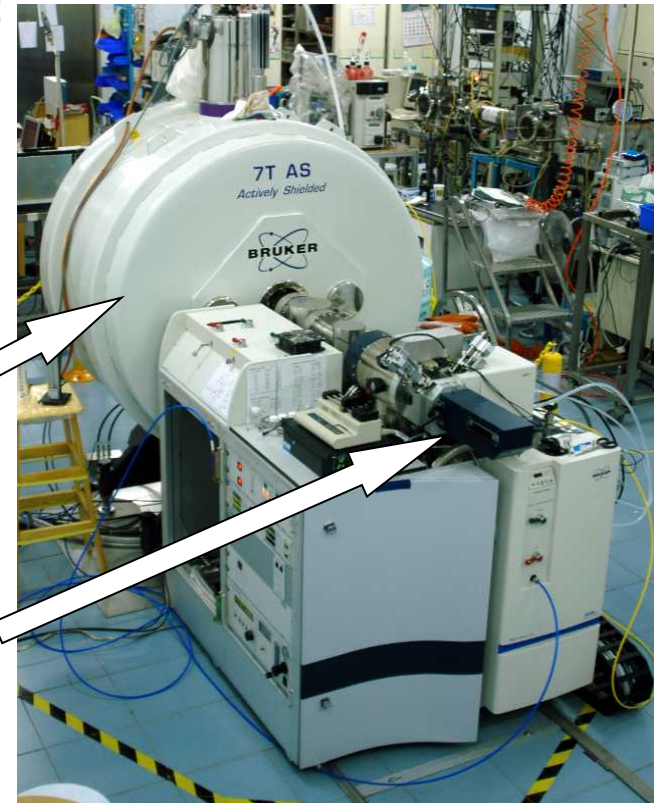
$$\omega_c = (q/m) \vec{B}$$

Configuration of Commercial FT-ICRMS



superconducting magnet

ion source



Anal. Chem. 72 (6), 1156 (2000).

Electric Potential

$$U(r,z) = \frac{k}{2} \left(z^2 - \frac{r^2}{2} \right) + \frac{k}{2} (R_m)^2 \ln \left[\frac{r}{R_m} \right] + C$$

characteristic radius

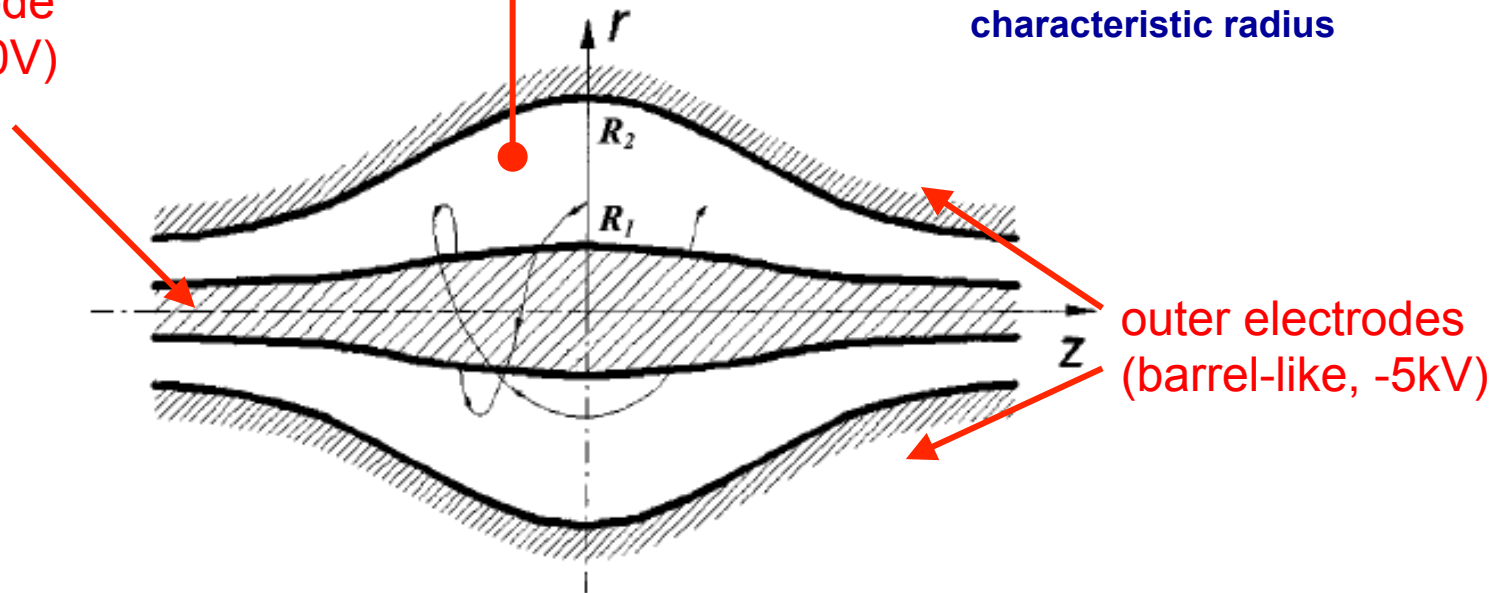


Figure 1. Equipotentials of the quadro-logarithmic field and an example of a stable ion trajectory

Fourier-Transform Orbitrap Mass Spectrometer

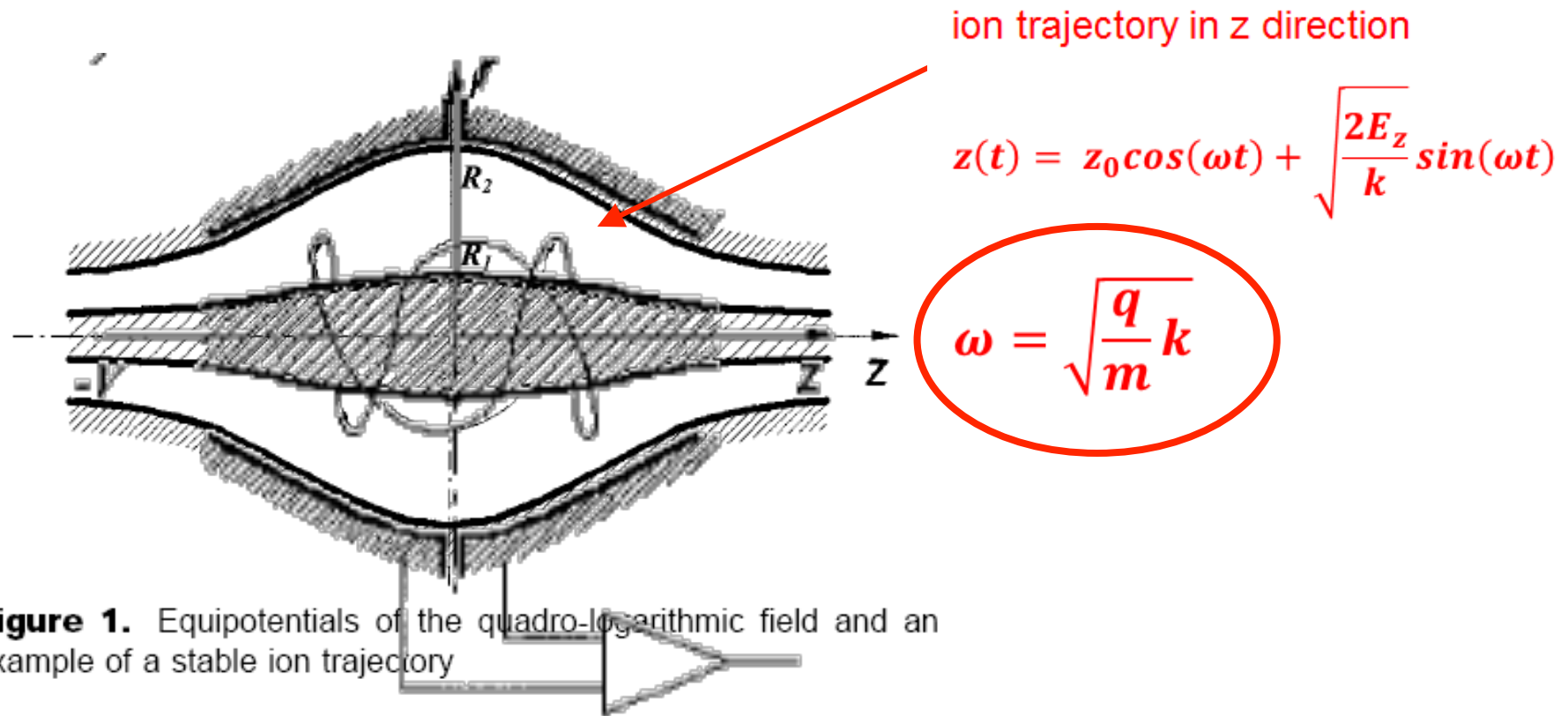
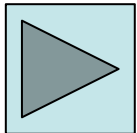


Figure 1. Equipotentials of the quadro-logarithmic field and an example of a stable ion trajectory



Comparison of Important Mass Analyzers

TABLE 3.1. Comparison of Various Types of Mass Analyzers

Characteristic	Magnetic	Quadrupole	QIT	LIT	TOF	FT-ICR	Orbitrap
Mass range (Da)	15,000	4000	4000	4000	Unlimited	$>10^4$	
Resolving power	10^2-10^5	4000	10^3-10^4	10^3-10^4	15,000	$>10^6$	150,000
Mass accuracy (PPM)	1-5	100	50-100	50-100	5-50	1-5	
Abundance sensitivity	10^6-10^9	10^4-10^6	10^3	10^3-10^5	up to 10^6	10^2-10^5	
Speed (Hz)	0.1-20	1-20	1-30	1-300	10^1-10^6	$10^{-2}-10^1$	
Efficiency%	<1	<1-95	<1-50	<1-99	1-100	<1-95	
Dynamic range	10^9	10^7	10^2-10^5	10^2-10^5	10^2-10^6	10^2-10^5	
MS/MS	Excellent	Great	Great	Excellent	Great	Great	
LC (CE)/MS	Poor	Excellent	Excellent	Excellent	Good	Good	
Cost	\$\$\$\$	\$	\$	\$\$	\$\$-\$\$\$	\$\$\$\$	

Heme

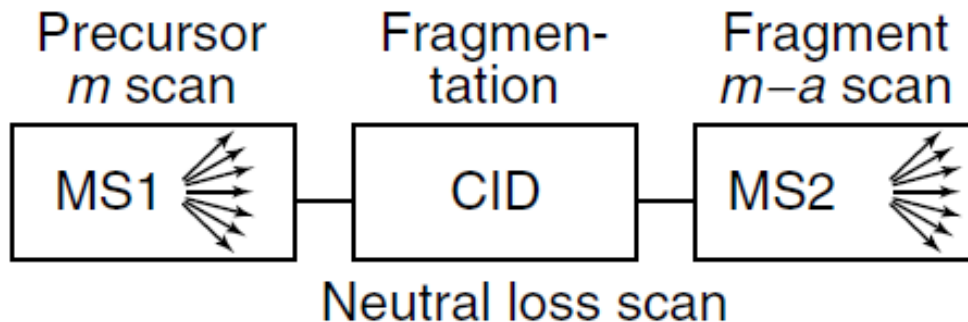
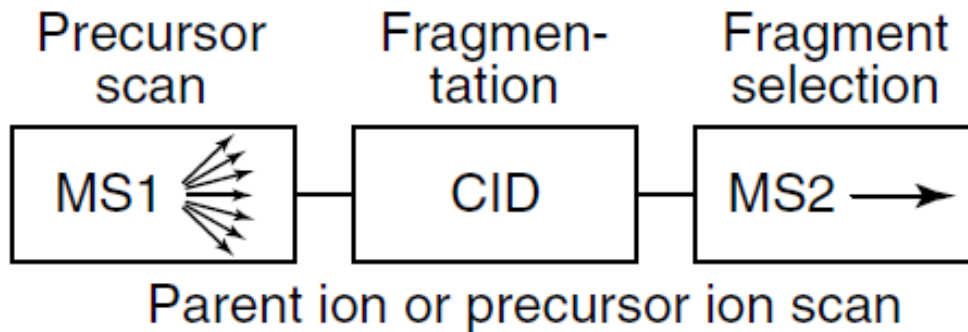
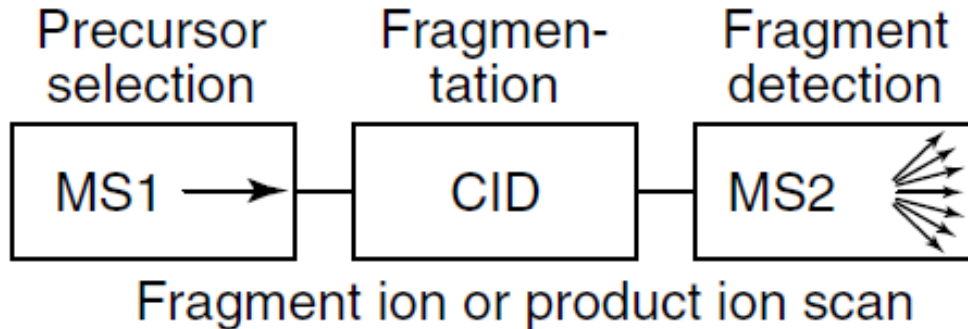
Fragmentation Methods

Table 1. General description of different ion activation processes presented

Method	Energy Range	Instruments	Description
PSD	Low	RETOF	Metastable or collision-induced dissociations in flight tube of reflectron time-of-flight instrument
CID	Low	QqQ, IT, QqTOF, QqLIT, FTICR	Collision-induced dissociation by collision of precursor ions with inert target gas molecules in collision cell. Energy range 1–100 eV
	High	Tandem TOF, sectors	Same as above with keV energies
SID	Low	Hybrid (BqQ), QqQ, IT, FTICR	Collisions between precursor ions and solid target surface with or without self-assembled monolayer causing fragmentations as well as other side reactions
	High	Tandem TOF, RETOF	Same as above with precursors of higher translational energies (instrument dependent)
ECD	Low	FTICR	Low-energy beam of electrons resulting in electron capture at
ETD (charge neutralization)		IT	protonation (or cationic) site with subsequent fragmentation following radical ion chemistry
IRMPD	Low	IT, FTICR	Continuous-wave low-energy infrared laser activates precursor ions by multiphoton absorption with consequent fragmentation
BIRD	Low	IT, FTICR	Low-energy thermal activation method ideal for calculations of energy thresholds and thermodynamic properties

Orbitrap

Fragmentation Methods



Sulfonation
SO₃: 80 Da

Phosphorylation
HPO₃: 80 Da
H₃PO₄: 98 Da

Fragmentation Technologies

(Bond-breaking process)

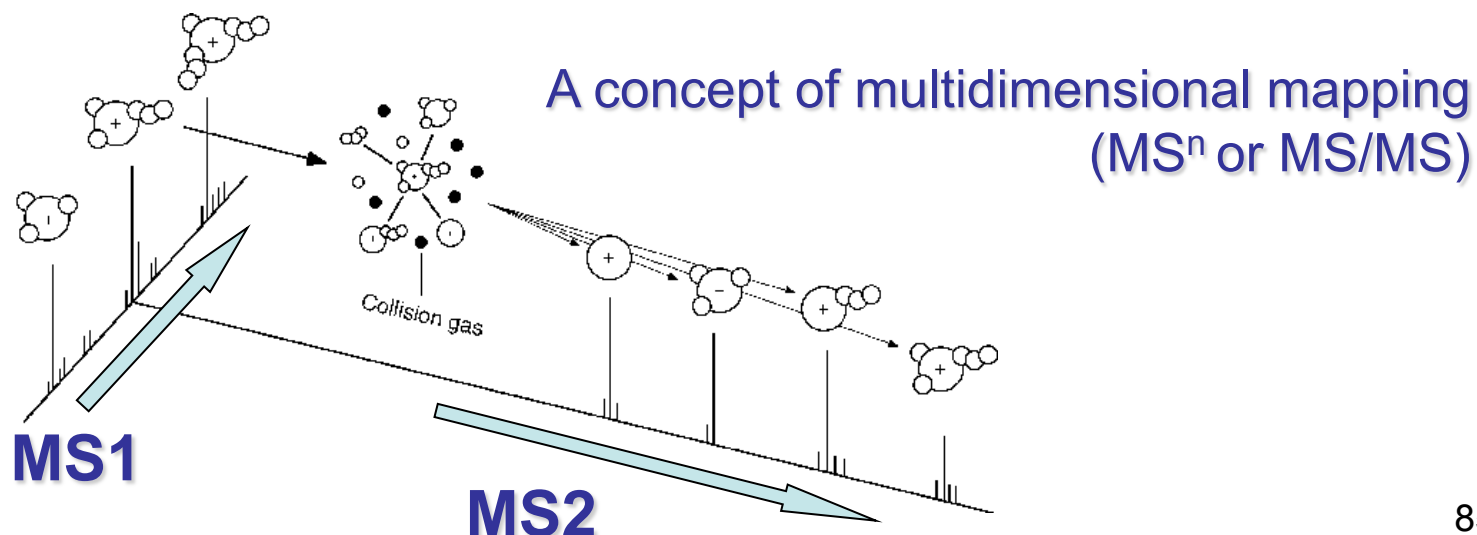


CID (collision-induced dissociation): Ion-neutral collision

IRMPD (infrared multiphoton dissociation): Laser heating

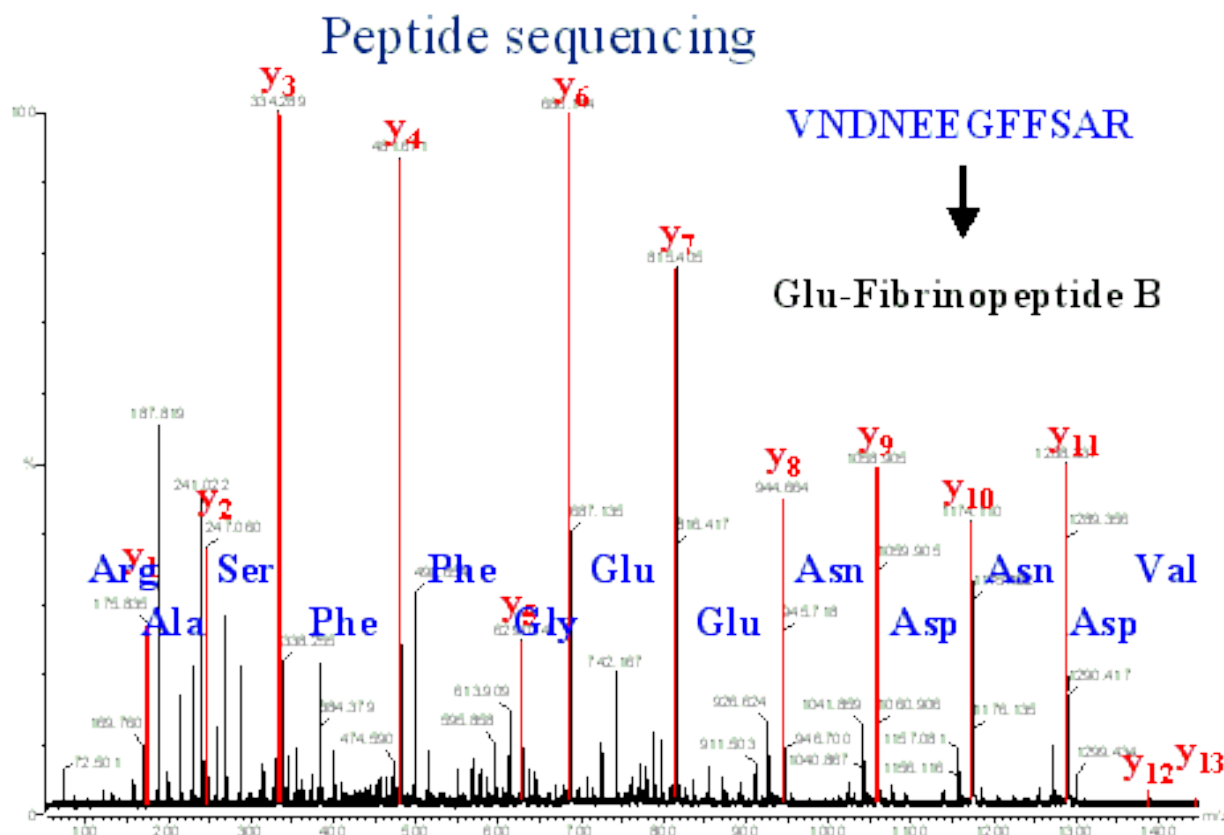
ECD (electron-capture dissociation): Charge recombination

ETD (electron-transfer dissociation): Charge recombination



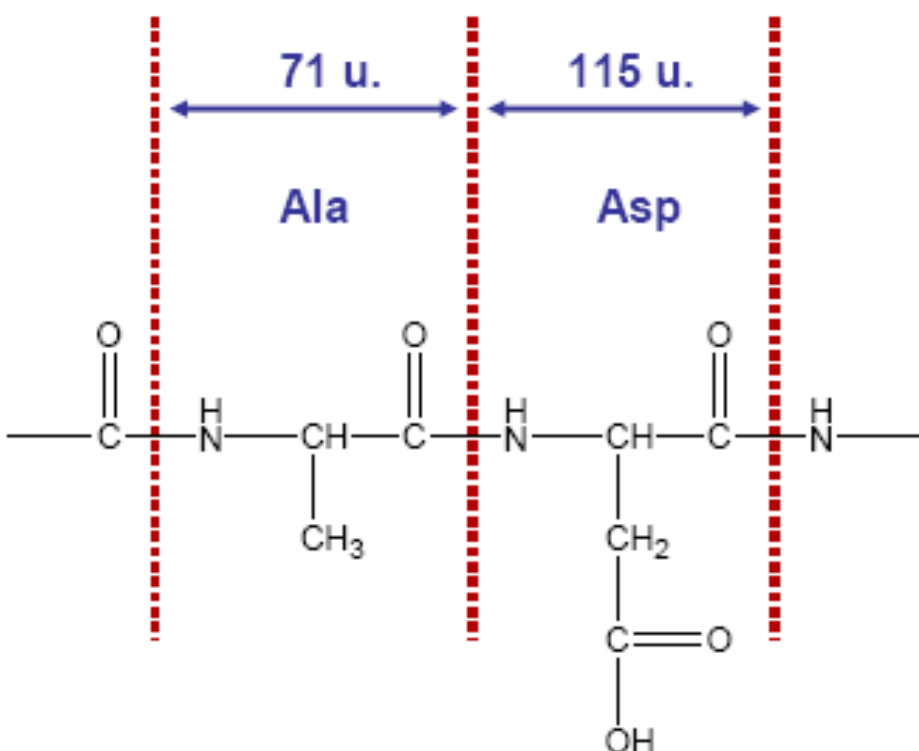
Peptide mass fingerprinting

Peptide Mass Fingerprinting (PMF) is a technique used to identify proteins by matching their constituent fragment masses (peptide masses) to the theoretical peptide masses generated from a protein or DNA database. This technique does well with 2D gel spots where the protein purity is high. PMF protein identification can run into difficulties with complex mixtures of proteins



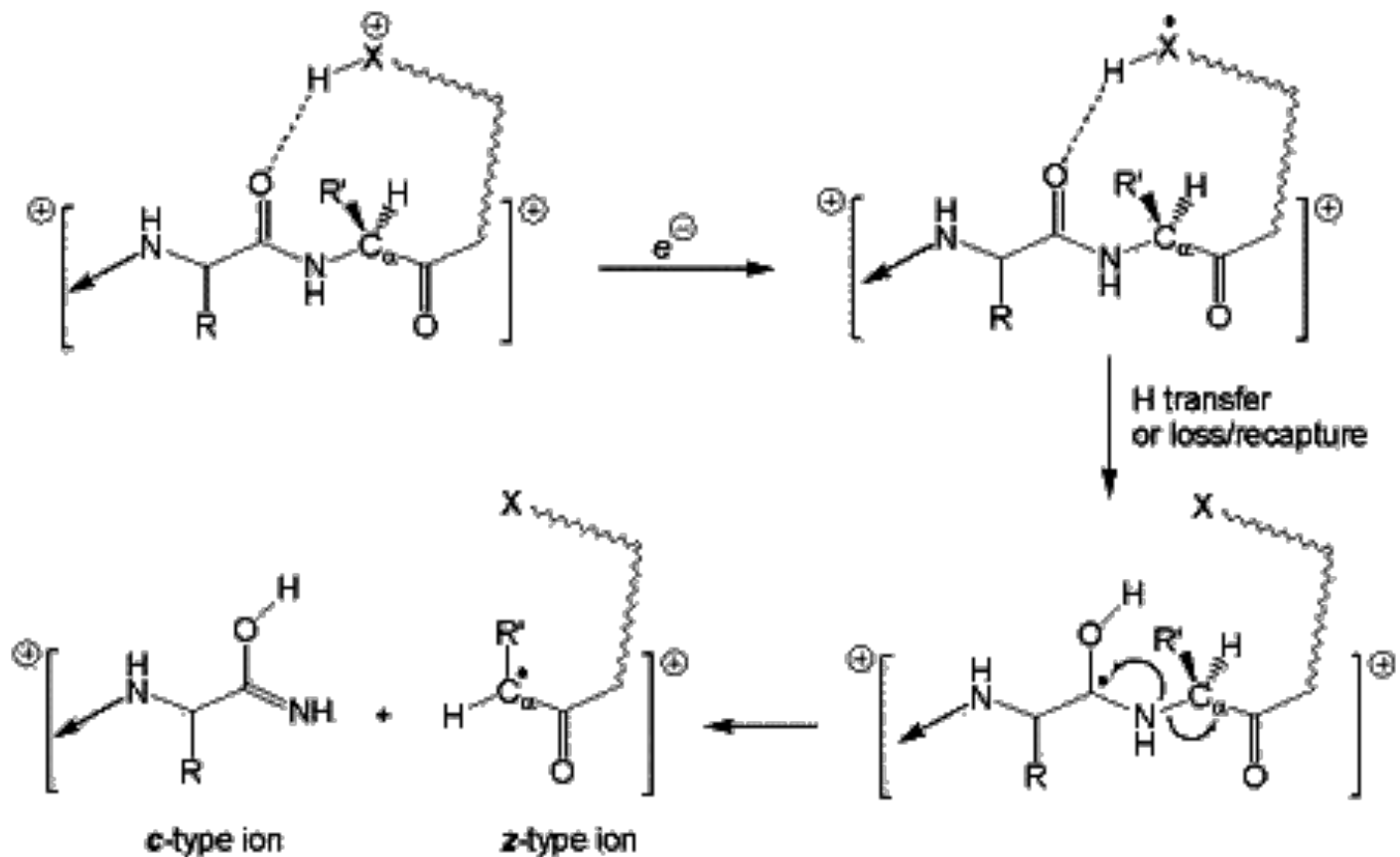
Dr H.M. Davey
Cledwyn Building
Institute of Biological Sciences
University of Wales

Peptide Sequencing



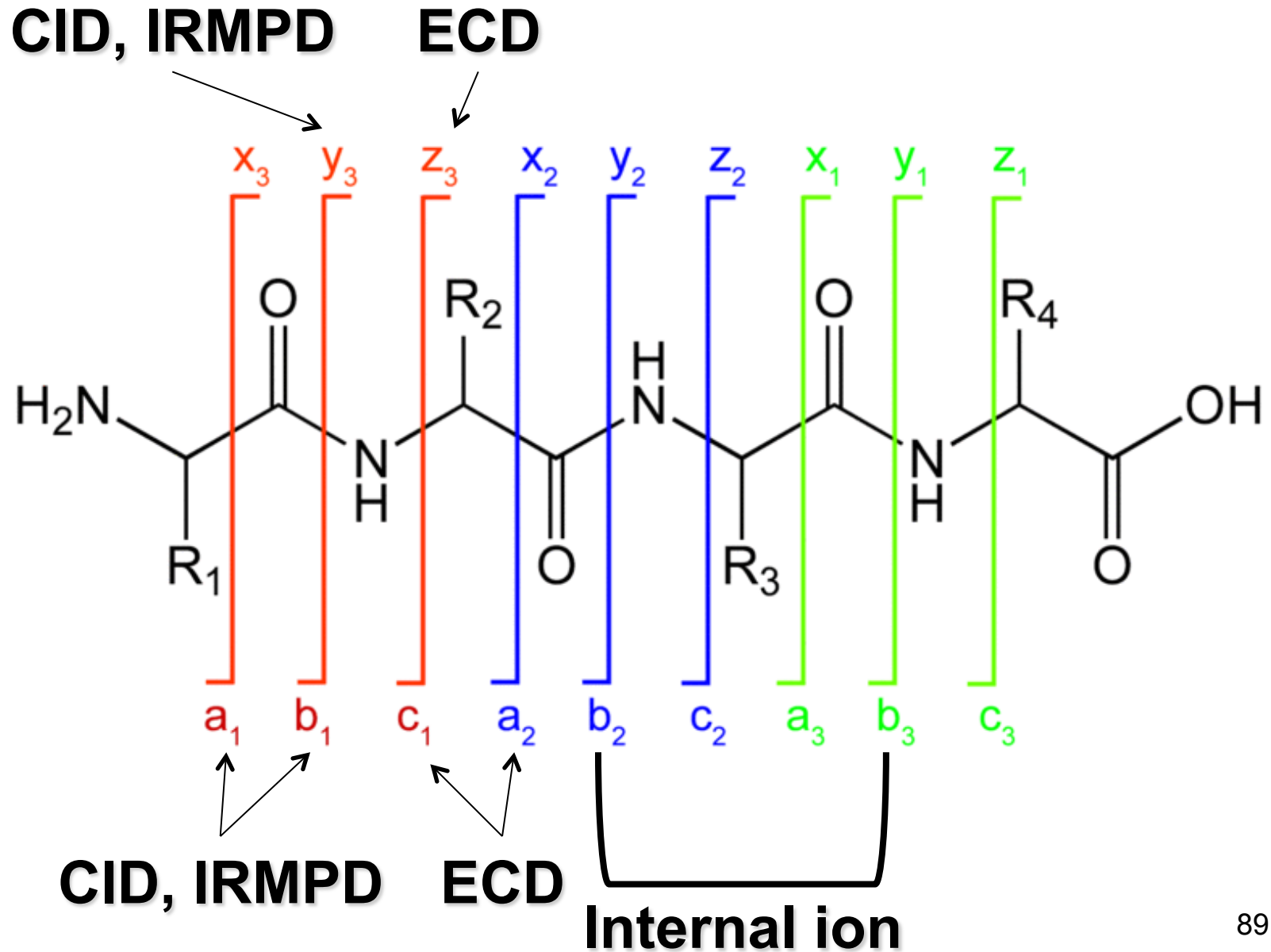
amino acid			mass
Alanine	ALA	A	71.09
Arginine	ARG	R	156.19
Aspartic Acid	ASP	D	115.09
Asparagine	ASN	N	114.11
Cysteine	CYS	C	103.15
Glutamic Acid	GLU	E	129.12
Glutamine	GLN	Q	128.14
Glycine	GLY	G	57.05
Histidine	HIS	H	137.14
Isoleucine	ILE	I	113.16
Leucine	LEU	L	113.16
Lysine	LYS	K	128.17
Methionine	MET	M	131.19
Phenylalanine	PHE	F	147.18
Proline	PRO	P	97.12
Serine	SER	S	87.08
Threonine	THR	T	101.11
Tryptophan	TRP	W	186.12
Tyrosine	TYR	Y	163.18
Valine	VAL	V	99.14

Electron-capture dissociation (ECD)

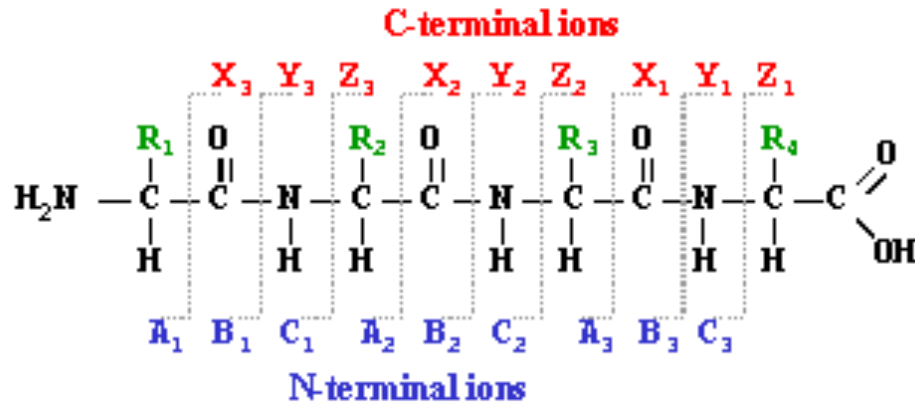


Scheme 1. The current ECD mechanism for H atom transfer and dissociation in internally-solvated peptide ions.

Cleavage Sites of Peptides



Site-specific cleavage on peptides: a/z, b/y, c/x fragmentations



P-A-F-D-S-I-M-A-E-T-L-K
(protonated mass 1410.6)

<u>mass⁺</u>	<u>b-ions</u>	<u>y-ions</u>	<u>mass⁺</u>
88.1	S	PAFDSIMAETLK	1323.6
185.2	SP	AFDSIMAETLK	1226.4
256.3	SPA	FDSIMAETLK	1155.4
403.5	SPAF	DSIMAETLK	1008.2
518.5	SPAFD	SIMAETLK	893.1
605.6	SPAFDS	IMAETLK	806.0
718.8	SPAFDSI	MAETLK	692.3
850.0	SPAFDSIM	AETLK	561.7
921.1	SPAFDSIMA	ETLK	490.6
1050.2	SPAFDSIMAE	TLK	361.5
1151.3	SPAFDSIMAET	LK	260.4
1264.4	SPAFDSIMAETL	K	147.2

Techniques for Protein Sequencing

Bottom-Up

This term refers to approaches that have been developed for the identification of proteins and the in-depth characterization of their structure. The term Bottom-Up reflects the **reconstruction of the primary structure of the proteins from the little pieces of sequences of the peptides** that could either be identified in databases or derived from the analysis of their mass spectra. Bottom-up approaches are often applied on protein mixtures in comparative proteomics.

Top-Down

This term refers to approaches that **attempt to determine the structure of the protein directly on the protein itself without breaking it into pieces by digestion**. The Top-Down approach makes use of mass spectrometers capable of isolating ions of the protein and breaking them into fragments of interpretable size that bear sequence information. The Top-Down approach relies also heavily on specific database-search engines to facilitate the interpretation.

Build up the protein model from peptide fragments.

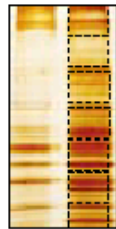
Bottom-Up Proteomics

1. Pre-separation

(1) Sample fractionation



SDS-PAGE



Excised proteins



(2) Trypsin digestion

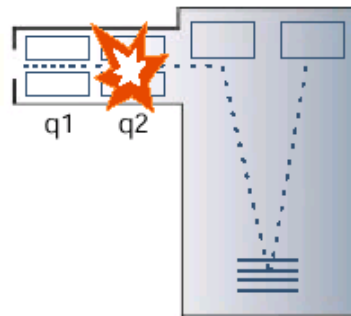


Peptide mixture



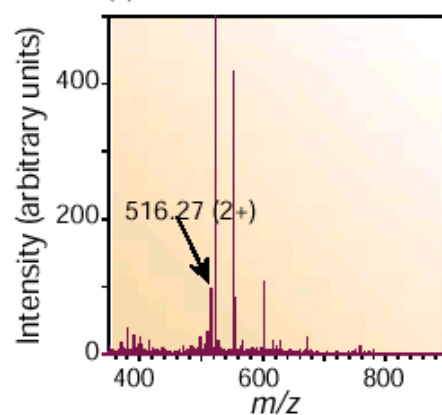
2. Digestion

(3) Peptide chromatography and ESI

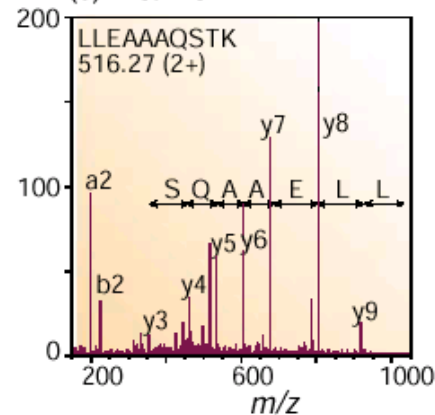


3. Mass detection

(4) MS



(5) MS/MS



R. Aebersold, M. Mann, Nautre, 422, 198 (2003).

Break down the protein structure starting from an intact molecule.

Top-Down Proteomics

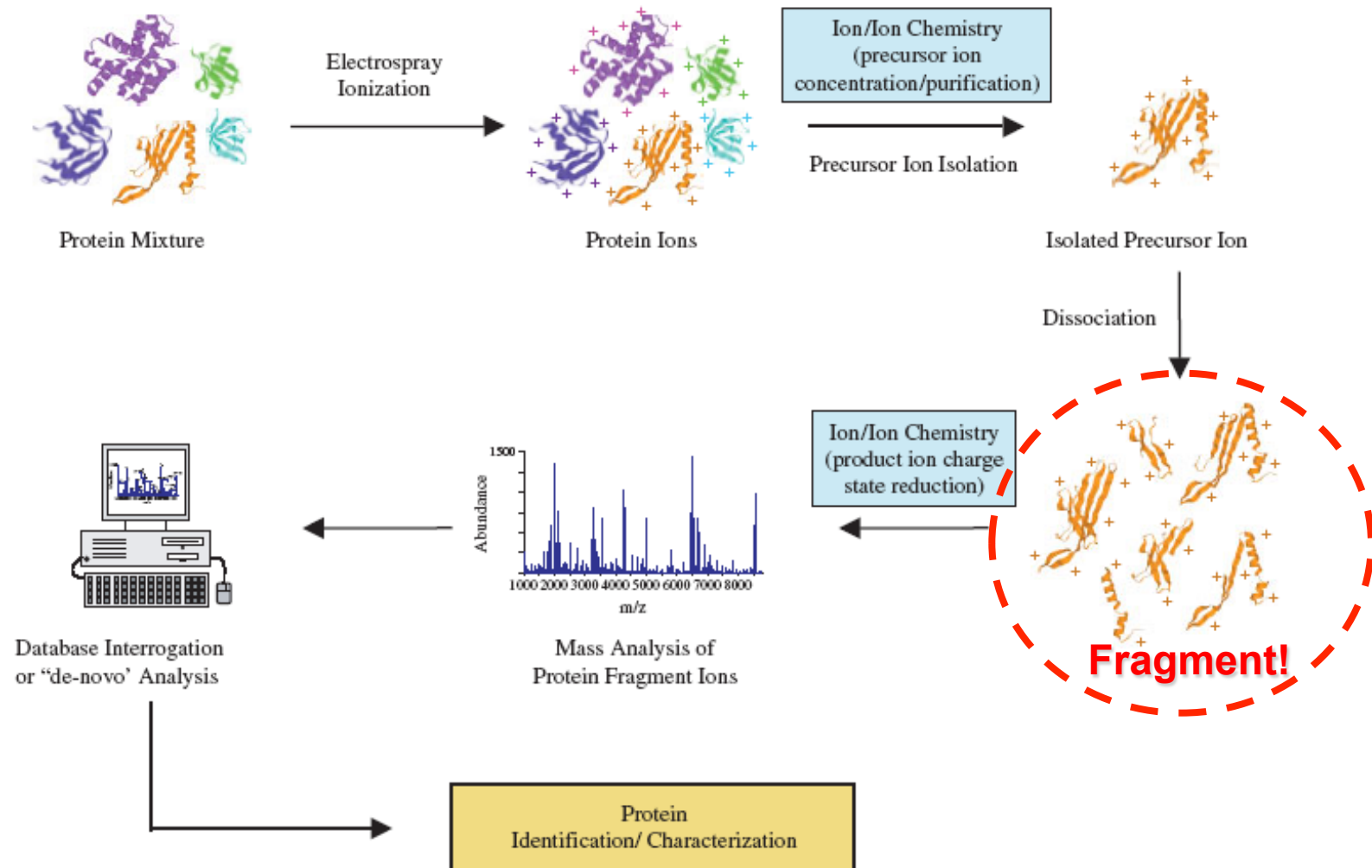


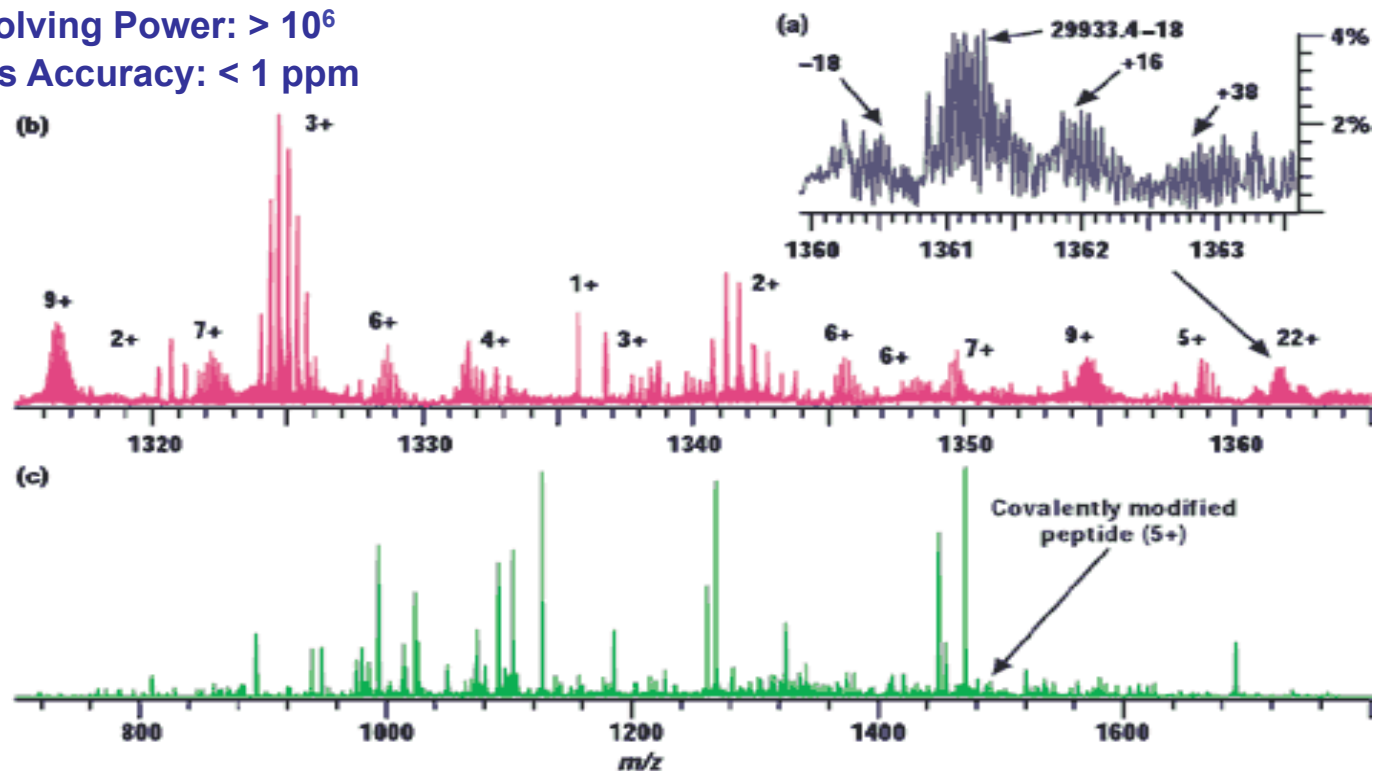
Plate 1. Schematic diagram of a typical top down protein characterization experiment. The light blue shaded boxes indicate where gas-phase ion-ion chemistries may be employed in top down approaches (see text).

WHEN

- *Identification of unknown molecules*
- *Imaging mass spectrometry*
- *Discussions and perspectives*

Advantage of High Resolution MS

Resolving Power: $> 10^6$
Mass Accuracy: < 1 ppm



(a, b) Two m/z scale-expanded segments from (c) an ESI FT-ICR broadband mass spectrum of a GluC digest of a 191-kDa protein.

The charge state of each peptide in (a) and (b) was determined from the reciprocal of the spacing between adjacent peaks in each isotopic distribution, of which there were 824. From the single broadband spectral data in (c), 581 individual peptides could be identified, along with the site of a single amino acid chemical modification.

Determining small molecule structure

Field	Res. (Mil.)	Acc. (ppm)
7T	>1	<1
9.4T	>1.5	<0.5
12T	>2	<0.3
15T	>2.5	<0.25

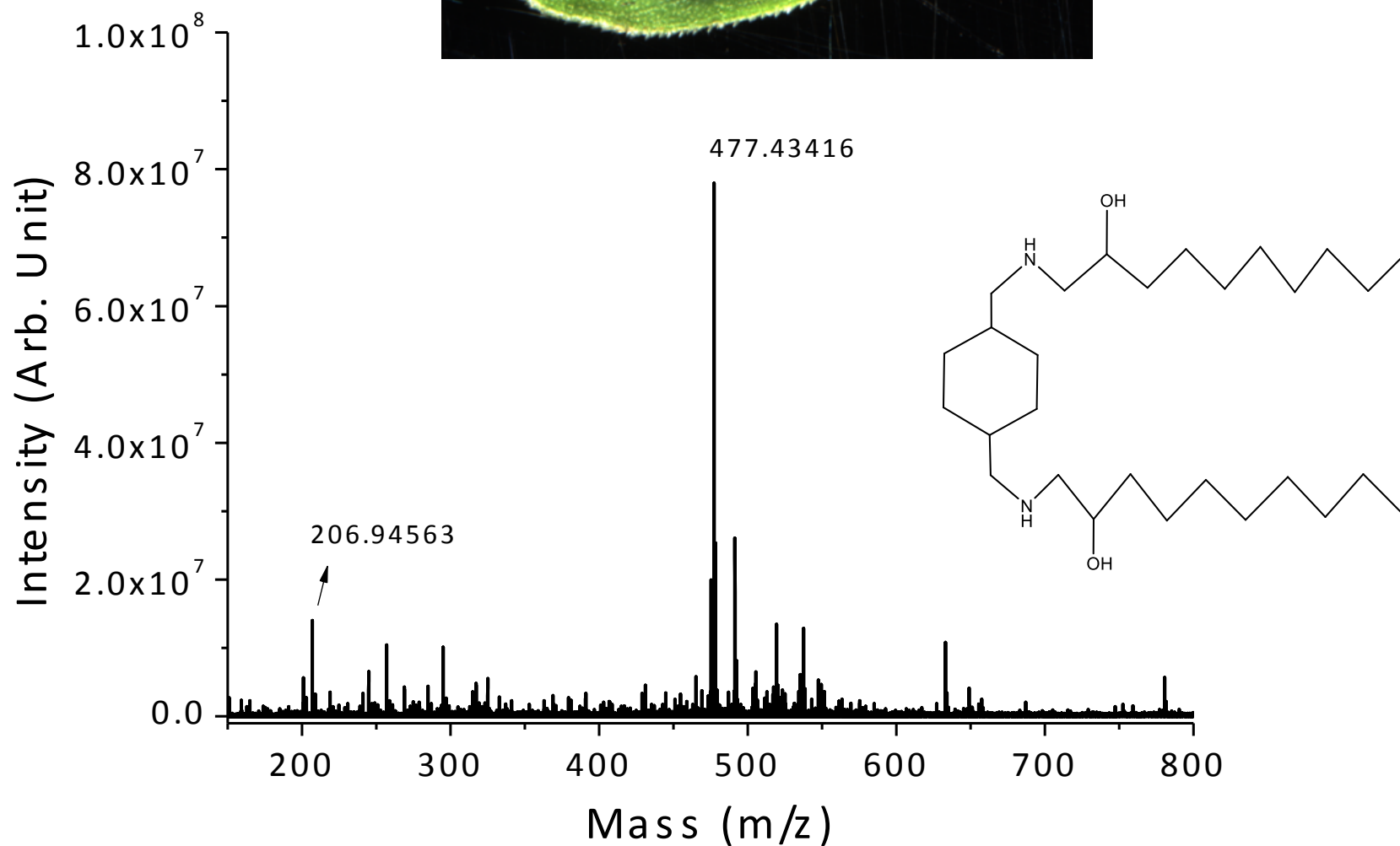
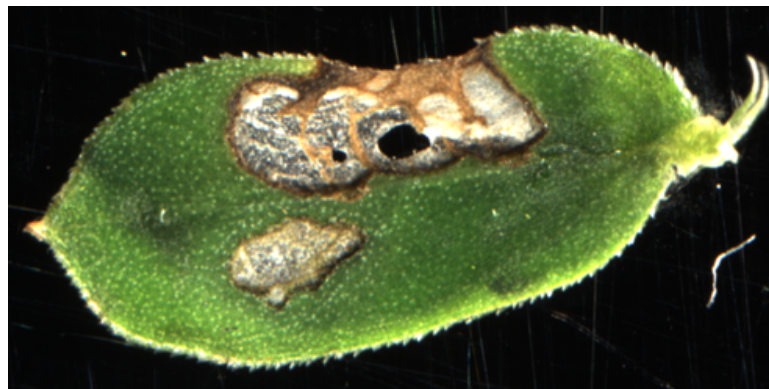
Bruker Daltonics' catalog

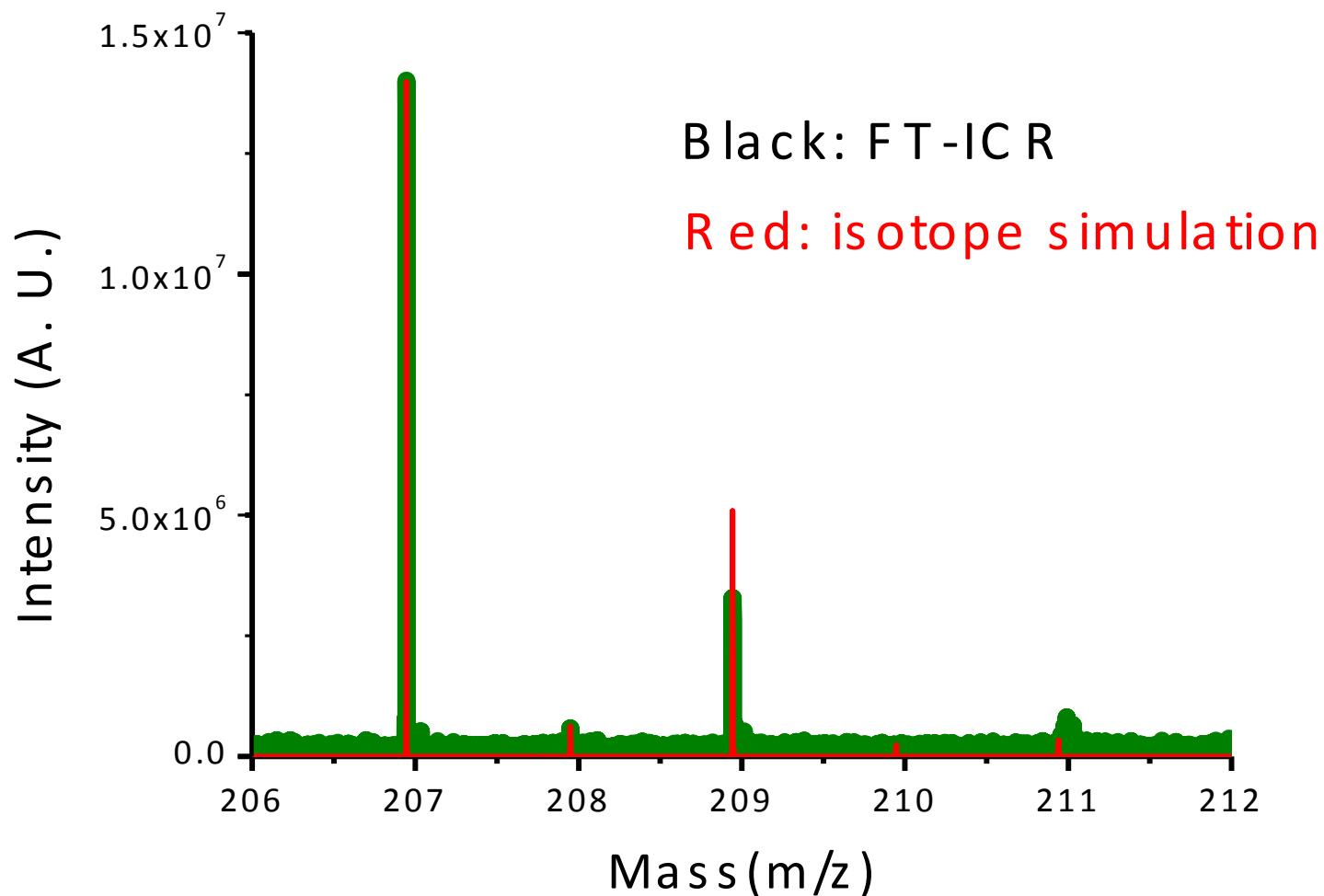
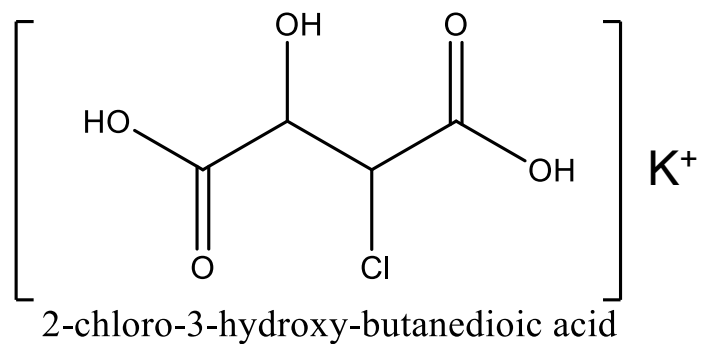
Measure the precise m/z,
FT-ICR (9.4T) mass accuracy < 0.5 ppm,
Resolving power > 1.5 M

**For M/Z = 1000,
 $\Delta M/Z < 0.001$**

Predict chemical formula
 $\Delta M/Z$, isotope pattern, π +ring, C/H ratio

Search structure on SciFinder
Search by chemical formula



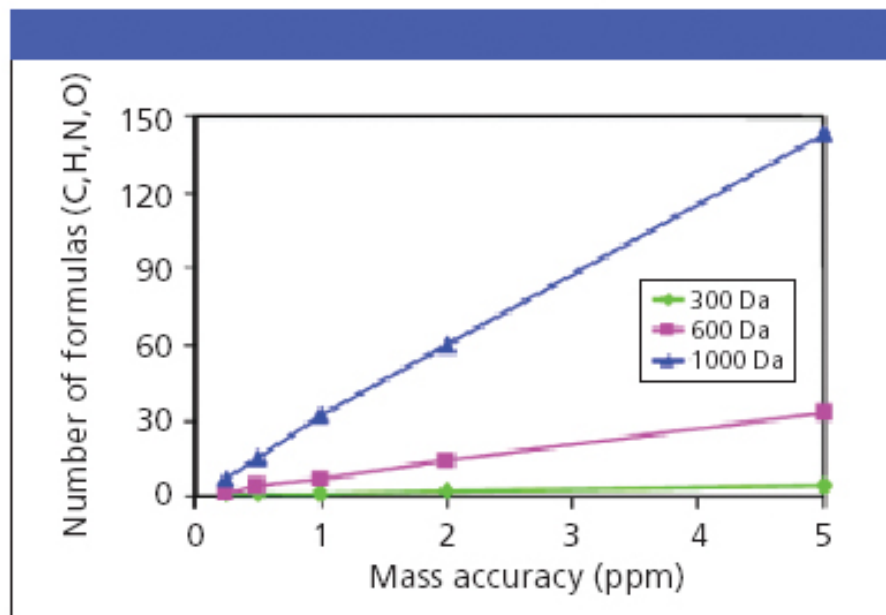


Measuring the precise m/z

For “unambiguous characterization”:

M/Z	Mass acc.
118	34 ppm
750	0.018ppm

Author’s guideline of the JASMS,
(March 2004)



T.L. Quenzer, et. Al., Automated accurate mass analysis using FTICR mass spectrometry, Proceedings of the 50th Annual Conference on Mass Spectrometry and Allied Topics, Orlando, Florida, 2002

Table 1: A comparison of leading higher resolution mass spectrometers (adapted from reference 1).

	Resolution (FWHM)	Mass range (m/z)	Mass accuracy (ppm)	Price (\$, typical range comparison)
TOF			2–5 typical	200–250K
Waters LCT Premier	10 000	18 000		
Agilent LC/MSD TOF	10 000	7 000		
Bruker MicroTOF	10 000	3 000		
Q-TOF			Same as TOF	300–700K (450K typical)
Waters QTOF Ultima	17 500	32 000		
Waters QTOF Micro	5 000	20 000		
MDS Sciex Qstar XL	10 000	40 000 (TOF) 20 000 (Q)		
Bruker BioTOF	20 000	10 000 (TOF)		
ICR (FT-MS)				500K–1.4M
Bruker Apex IV	100 000	66 000	<1	
Ion Spec	1300 000	18 000	1	
Thermo LTQ FT	500 000	2 000	2	

Debating resolution and mass accuracy,
Micheal P. Balogh, 2004

Predict chemical formula

SmartFormula Manually

Min: K Cl
Max: K
Cl 1-n, K 1-1

Note: for $m < 2000$ the elements C, H, N, and O are considered implicitly.

Measured m/z : 206.94563 Tolerance: 20 ppm Charge: 1

#	Mol. Formula	m/z	err [mDa]	err [ppm]	err [ppm]	mean err [ppm]	mSigma	Sigma Rank
1	C ₅ HClKN ₄ O	206.94705	1.42	6.8	6.8	5.4	73.2	1
2	C ₂ H ₃ ClKN ₃ O ₄	206.94437	-1.26	6.1	-6.1	-7.5	74.9	2
3	C ₄ H ₅ ClK ₂ O ₅	206.94571	0.08	0.4	0.4	-1.0	75.8	3
4	C ₇ H ₃ ClKN ₂ O ₂	206.94839	2.76	13.3	13.3	11.9	76.1	4
5	C ₄ H ₆ Cl ₂ KN ₂ O	206.94888	3.25	15.7	15.7	14.1	187.4	5
6	C ₂ H ₄ Cl ₂ KN ₅	206.94753	1.90	9.2	9.2	7.7	202.2	6

☐ Automatically locate monoisotopic peak
☒ Check rings plus double bonds
☒ Filter H/C element ratio
☒ Estimate carbon number

Maximum number of formulas: 500
 Minimum: -0.5 Maximum: 40
 Electron configuration: both
 Minimum H/C: 0 Maximum H/C: 3
☐ Generate immediately

Show Pattern

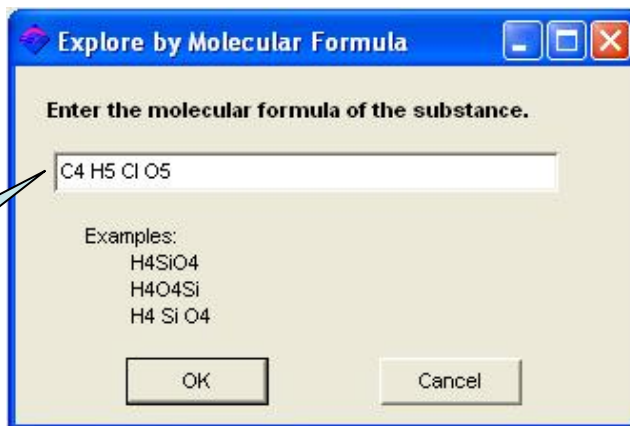
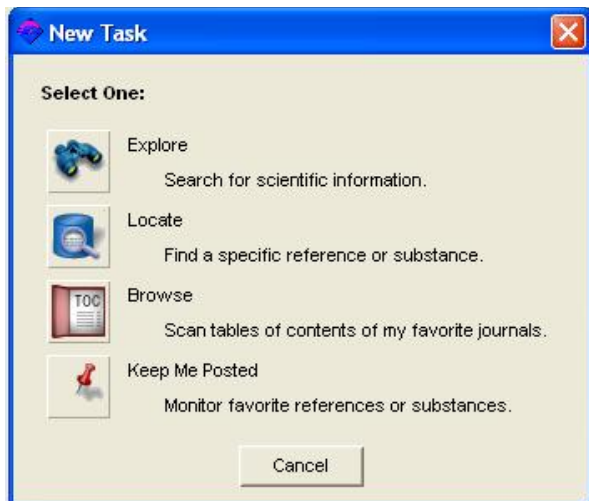
Atoms that must / must not appear

M/Z bias

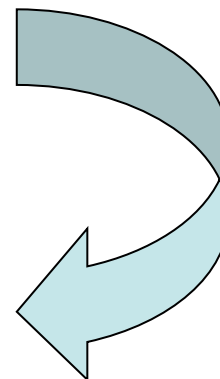
Isotope pattern similarity

Feature of the molecule. Ex.
 Carbohydrate ~1:2
 Protein 1:1.4~1.5
 DNA ~1:1.2

Search structure on SciFinder



Note that charge carrier (K^+) was removed



SciFinder

File Edit View Task Tools Help

New Task Back Forward Print Save As Full Text Prefs Combine History Contact Internet Patview Help Exit

<p>859980-78-2</p> <chem>O=C(O)C(O)C(O)C(=O)Cl</chem> <p>~1 Reference REGISTRY</p>	<p>742046-86-2</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>No References REGISTRY</p>	<p>165877-67-8</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~1 Reference REGISTRY</p>
<p>112335-64-5</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~1 Reference REGISTRY</p>	<p>74841-95-5</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~2 References REGISTRY</p>	<p>74804-17-4</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~1 Reference REGISTRY</p>
<p>21788-47-6</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~4 References REGISTRY</p>	<p>19071-25-1</p> <chem>O=C(O)C(O)C(Cl)C(=O)O</chem> <p>~12 References REGISTRY</p>	<p>7403-74-9</p> <chem>O=C(O)C(Cl)C(O)C(=O)O</chem> <p>~2 References REGISTRY</p>

Get References Get Reactions Analyze/Refine Back

Substances 1-9 of 9

Imaging Mass Spectrometry

making mass spectrometer the chemical microscope

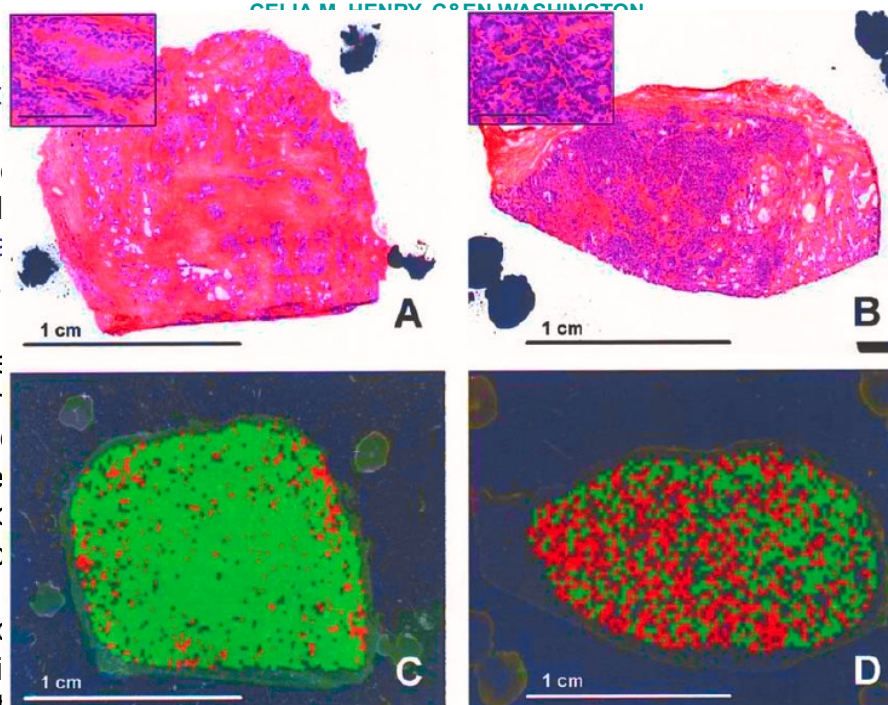
DRAWING WITH MASS SPEC

Mass spectrometry is emerging as a tool to image biological samples from single cells to brain slices

If a picture is worth a thousand words, mass spectrometric imaging takes familiar mass spectrometry and adds a two-dimensional sample. Especially in biological samples, the information associated with fields of view is

These pictures are acquired by mass spectrometry. Components can be represented as "pixels." Multiple pictures are taken, each smaller, more tightly focused, to create an image that takes longer to create.

Most mass spectrometry is assisted laser desorption/ionization (MALDI). The sample is covered with a matrix. The matrix molecules are ionized by the laser, and the sample molecules are ionized by the matrix. In SIMS, the sample is ionized by a primary ion beam, and the sample molecules are ionized by the primary ion beam. The two methods are complementary: MALDI works well for relatively large molecules such as peptides and proteins, whereas SIMS is better for small molecules.



al information? Mass spectrometry provides the "colors." It is a microscope. So, given a mass spectrometry, masses are located. Visual inspection long

n or the sample, and the distribution of specific components at each spot, or a set of spectra. Using mass spectrometry means that the

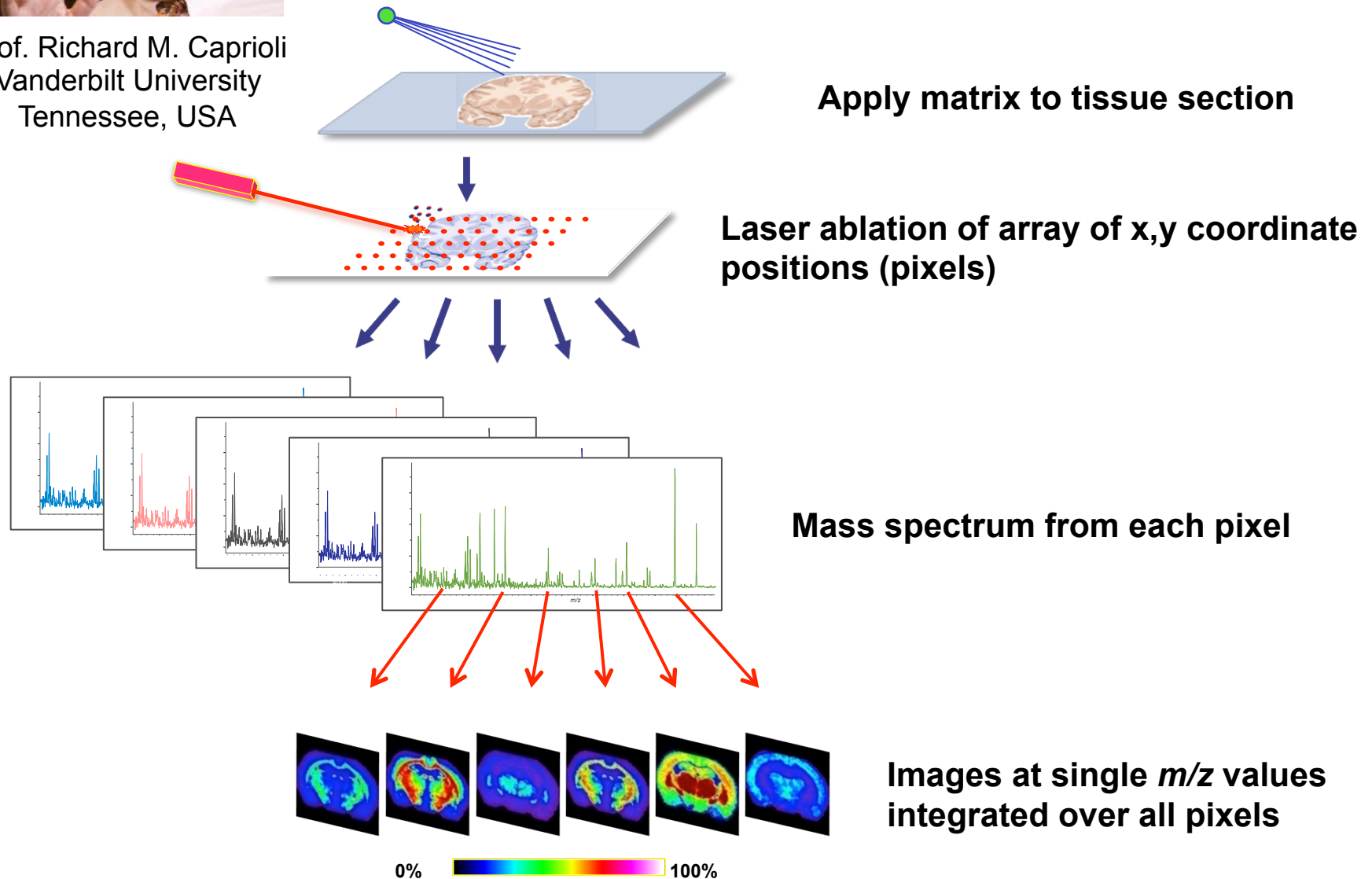
n the sample: matrix-assisted laser desorption/ionization (MALDI). In MALDI, the sample is desorbed and ionized by the matrix. The sample molecules are ionized by the matrix. The sample molecules are ionized by the matrix. The sample molecules are ionized by the matrix.

[Schwamborn K. et al, Int. J. Mol. Med. \(2007\), 20, 155-159.](#)



Prof. Richard M. Caprioli
Vanderbilt University
Tennessee, USA

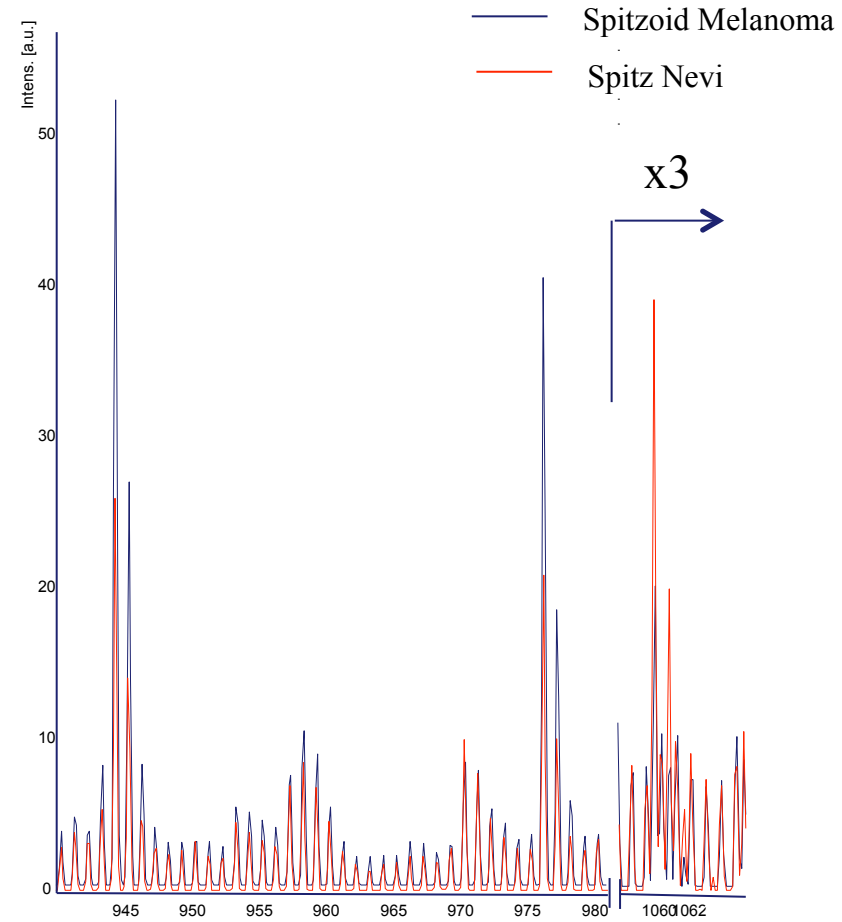
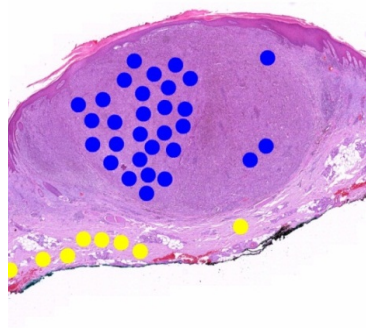
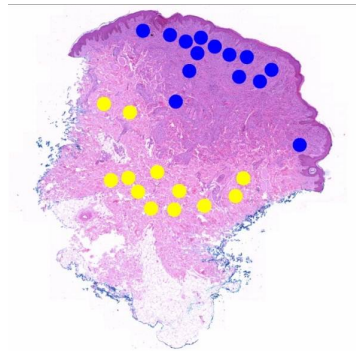
Basic Approach of MALDI Imaging MS for Tissue Analysis





Prof. Richard M. Caprioli
Vanderbilt University
Tennessee, USA

MS Analysis of Spitzoid Lesions in FFPE Biopsies



Erin Seeley, Rossitza Lazova (Yale)



Prof. Richard M. Caprioli
Vanderbilt University
Tennessee, USA

Classification of Spitzoid Lesions

Training set	# Patients	Classification Accuracy (%)
Spitz nevi (SN)	26	100
Spitzoid Malignant Melanoma (SMM)	25	96

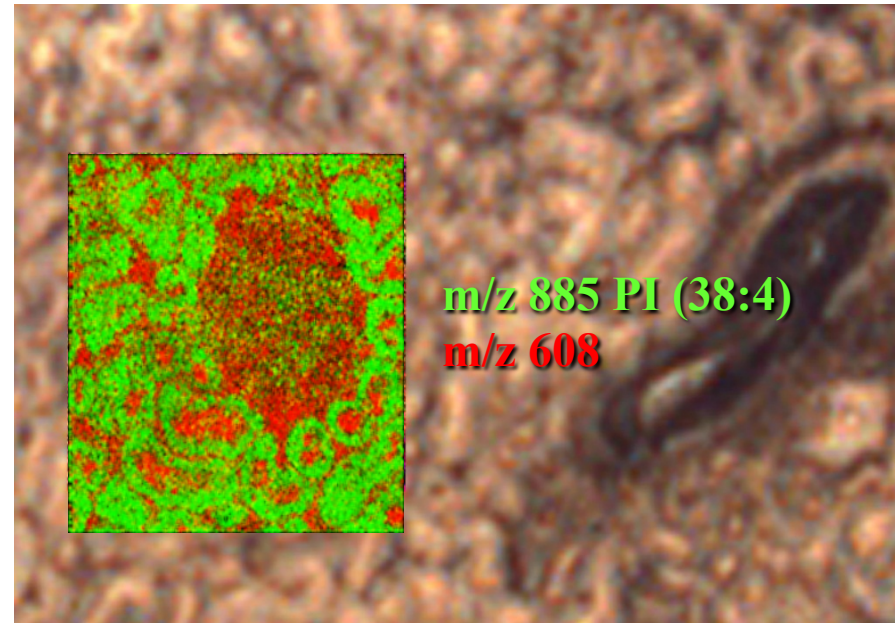
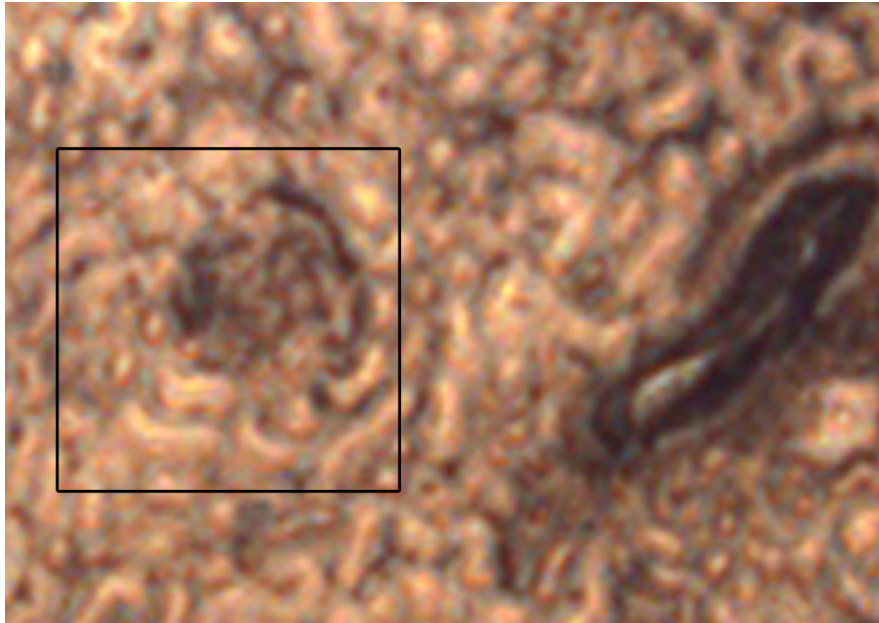
Validation (test) set	# Patients	Classification Accuracy (%)
Spitz nevi (SN)	30	97
Spitzoid Malignant Melanoma (SMM)	29	90



Prof. Richard M. Caprioli
Vanderbilt University
Tennessee, USA

Ion Images of Human Kidney Cortex

Resolution: 1 μm laser beam, 2 μm pitch
25 shots/pixel, transmission geometry
matrix sublimed DAN

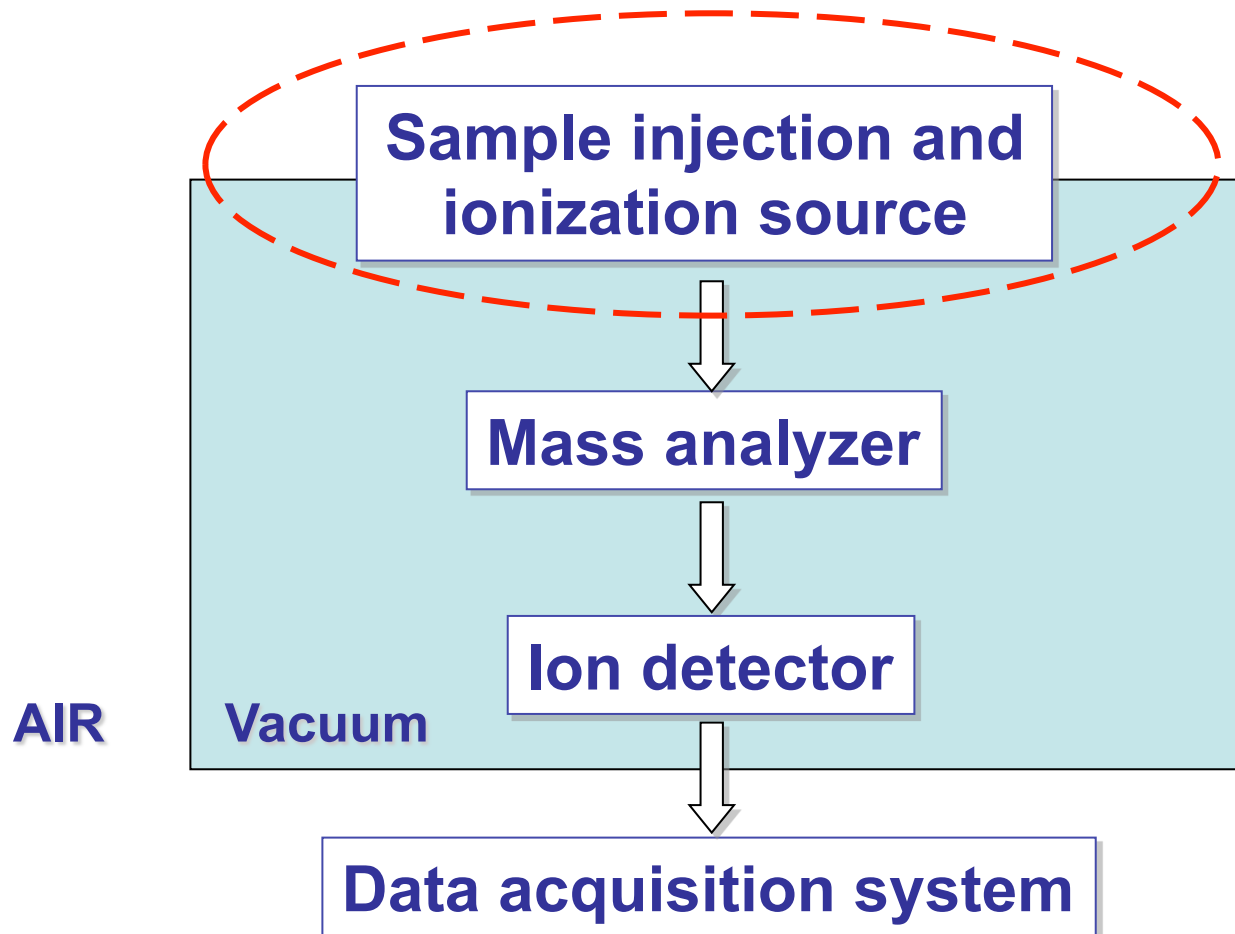


m/z 750 PE (18:0p/20:4)

m/z 1052 Hex-Sulfo-Hex-Cer-
(d16:1/26:0)

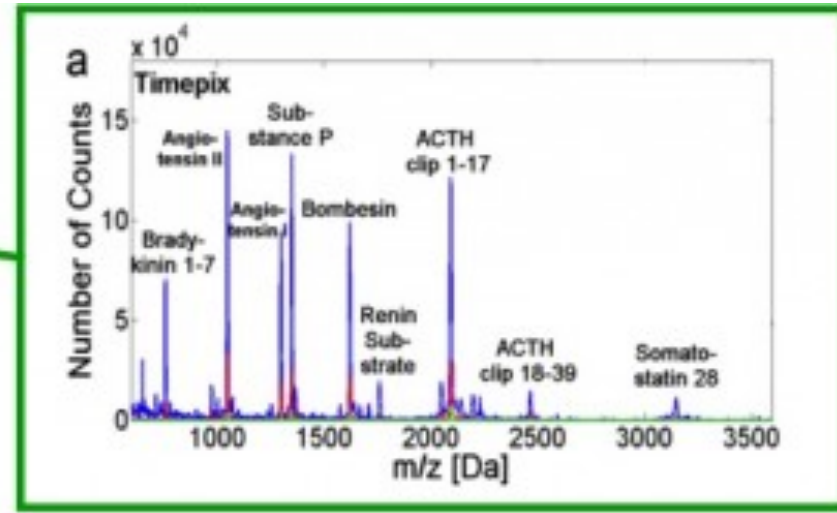
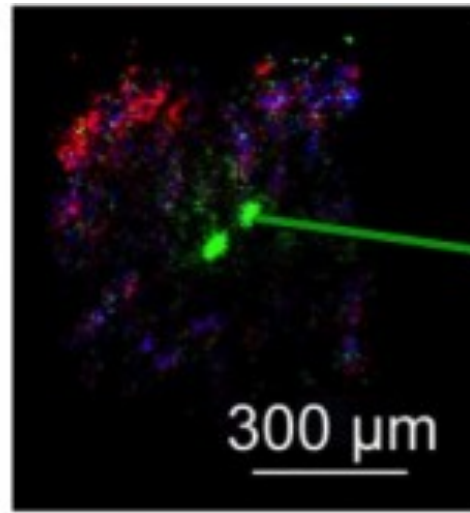
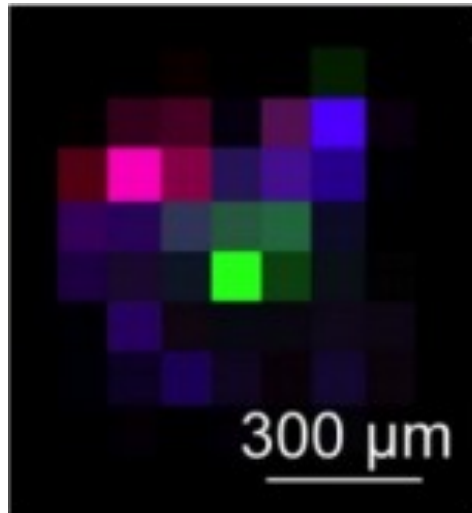
m/z 906 SulfoHex-Cer (t18:0/24:1)

General Configuration of Mass Spectrometer



Available Imaging Techniques

- Millimeter scale imaging: DESI
- Micron scale imaging: MALDI
- Sub-micron scale imaging: SIMS



OMICS2IMAGE

Imaging mass spectrometry in millimeter resolution

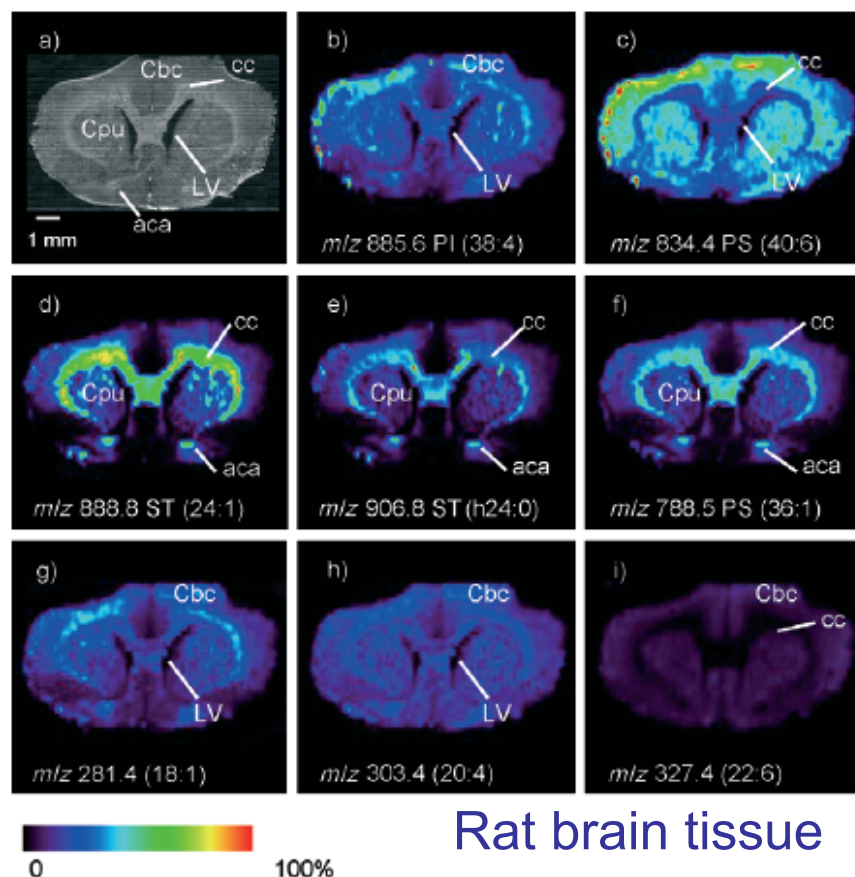
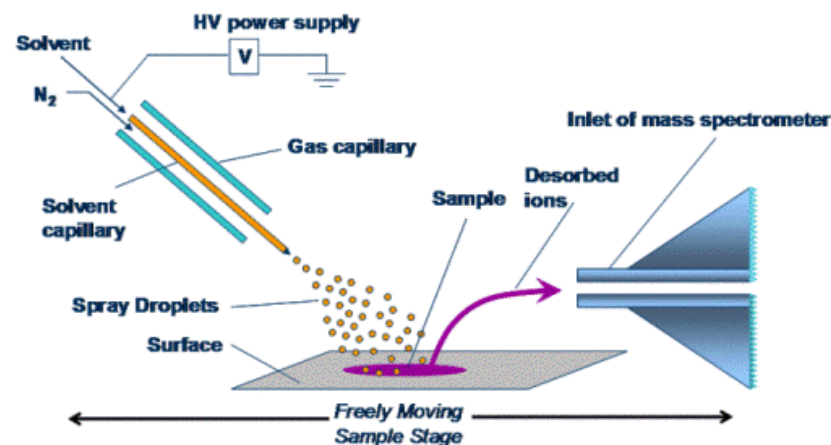
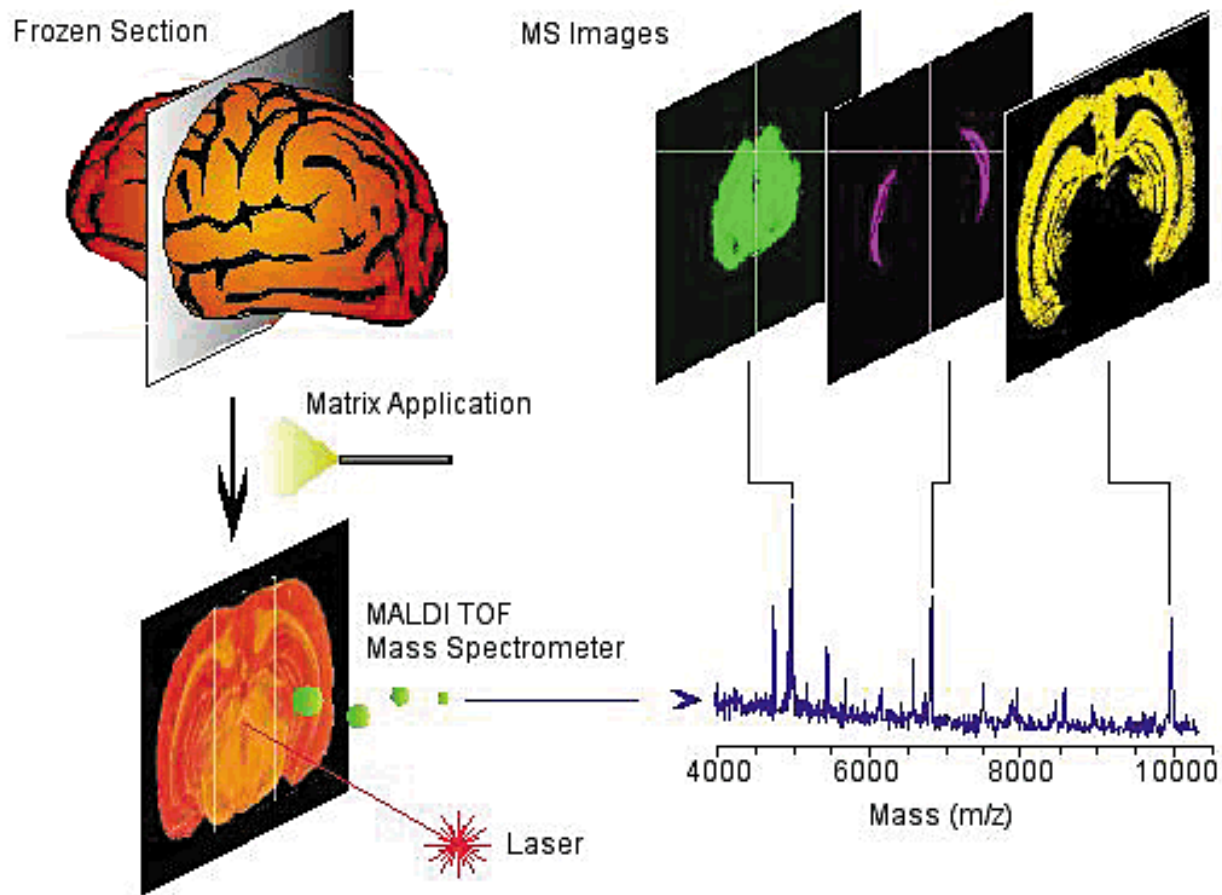


Figure 2. Selected molecular ion $[M-H]^-$ images of specific lipids from analysis of a $13 \times 10 \text{ mm}^2$ area of rat brain tissue section. a) Optical image of the coronal section of the rat brain prior to analysis. cc = corpus callosum; Cpu = striatum; Cbc = cerebral cortex; LV = lateral ventricle; aca = anterior part of anterior commissure. b-i) Ion images of PI (38:4); b), PS (40:6); c), ST (24:1); d), ST (h24:1); e), PS (36:1); f), oleate (18:1); g), arachidonate (20:4); h), and docosahexaenoate (22:6); i).

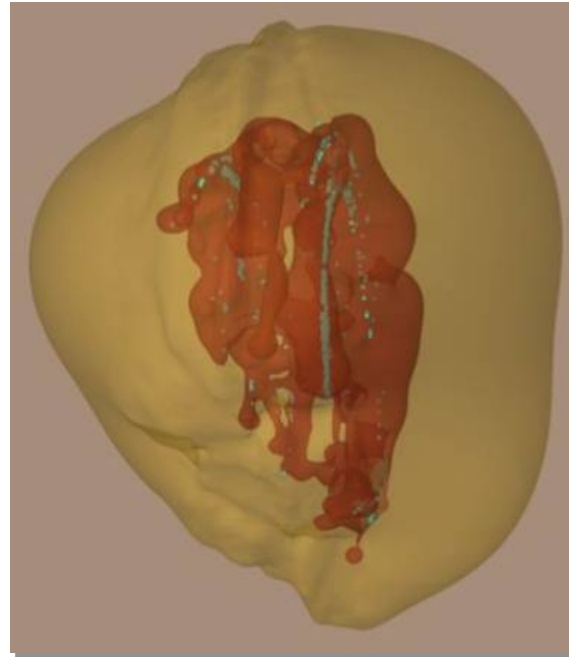
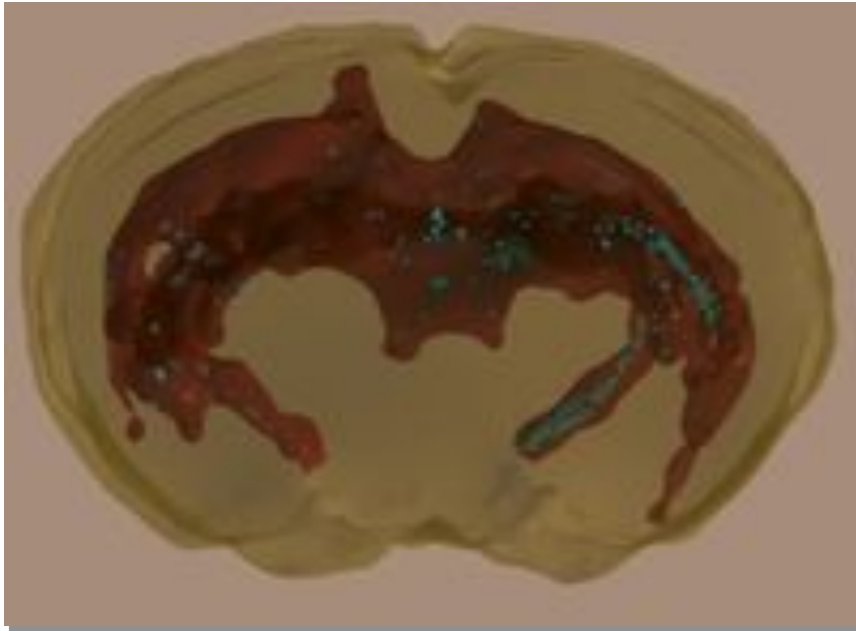


Prosolia Inc.

Imaging mass spectrometry in micrometer resolution

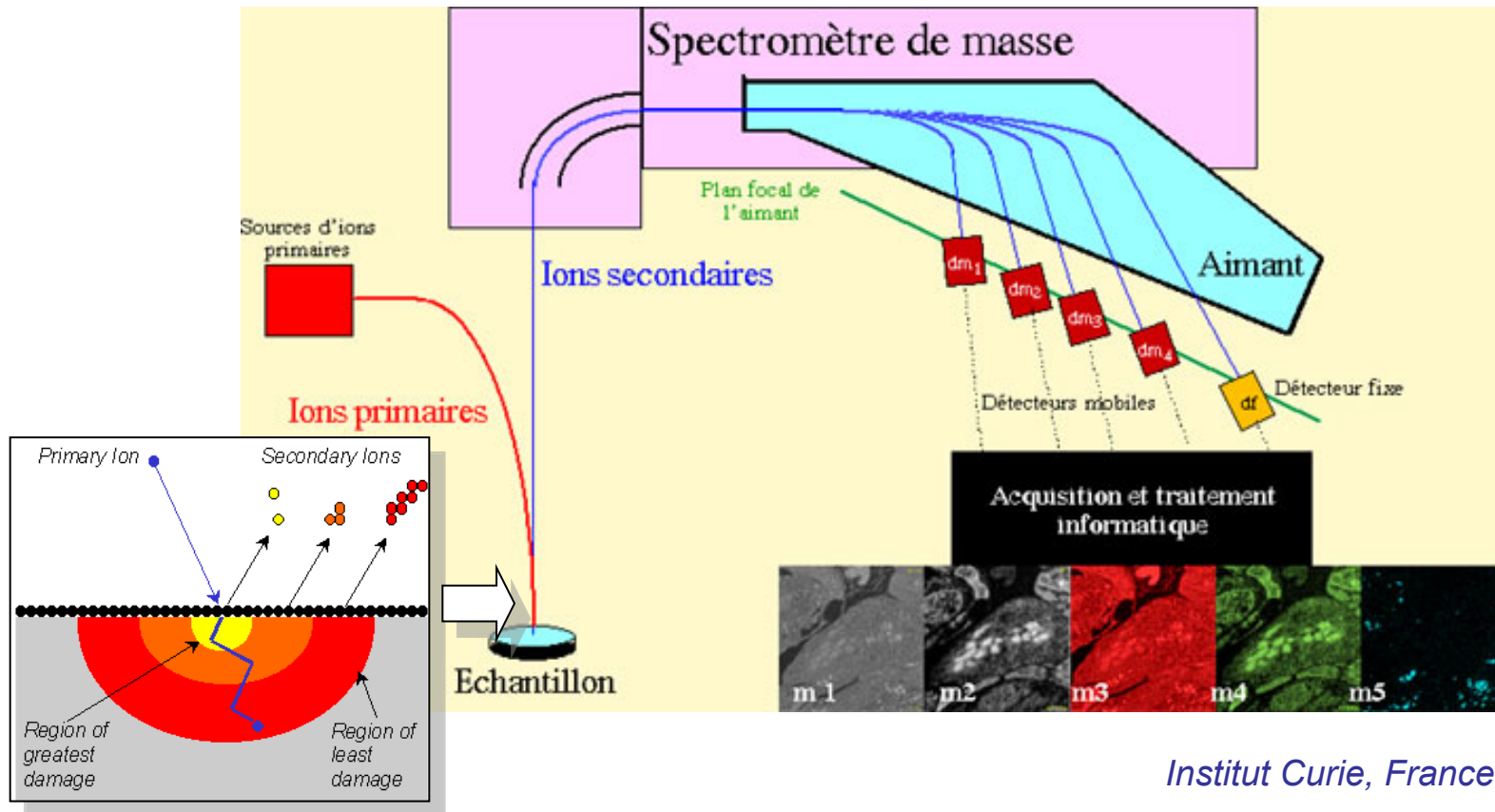


Computer assisted 3-D reconstruction of the corpus callosum



Bobby Bodenheimer, Department of Electrical Engineering and Computer Science at Vanderbilt University

Secondary ion mass spectrometer



Institut Curie, France

Physical Electronics

Imaging and differentiation of mouse embryo tissues by ToF-SIMS

Ligang Wu^{a,b}, Xiaochen Lu^b, Kristen S. Kulp^b, Mark G. Knize^b, Elena S.F. Berman^b,
Erik J. Nelson^b, James S. Felton^{a,b}, Kuang Jen J. Wu^{b,*}

^a Department of Applied Science, University of California-Davis, Davis, CA 95616, United States

^b Chemistry, Materials and Life Sciences Directorate, Lawrence Livermore National Laboratory, Livermore, CA 94551, United States

Received 1 June 2006; received in revised form 11 September 2006; accepted 11 September 2006

Abstract

Time-of-flight secondary ion mass spectrometry (ToF-SIMS) was used to differentiate tissue types (and repeatability of the method) by principal component analysis. Mouse embryos can be differentiated by ToF-SIMS to determine subtle chemical differences. Published by Elsevier B.V.

Keywords: ToF-SIMS; Mouse embryo

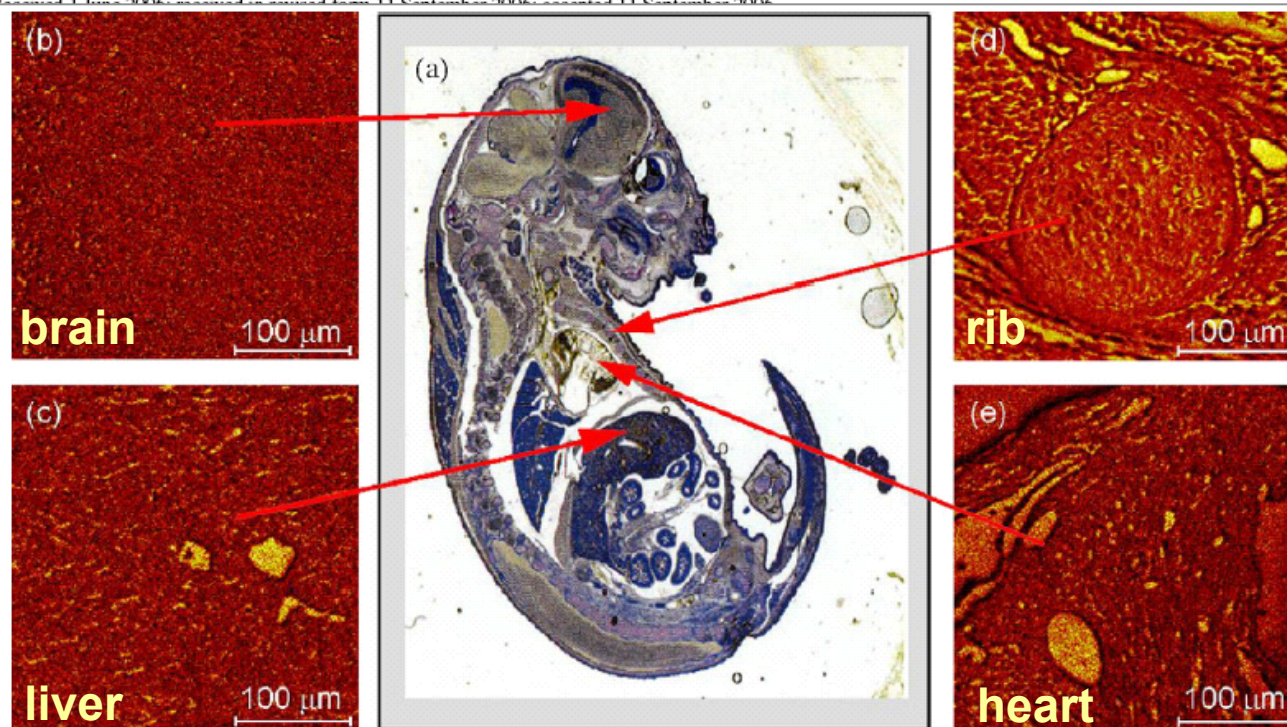


Fig. 1. (a) Optical image of a 16-day-old H&E stained mouse embryo section. (b) Positive total ion ToF-SIMS image of brain. (c) Positive total ion ToF-SIMS image of liver. (d) Positive total ion ToF-SIMS image of rib. (e) Positive total ion ToF-SIMS image of heart. Arrows point to the corresponding tissue region in the optical image.

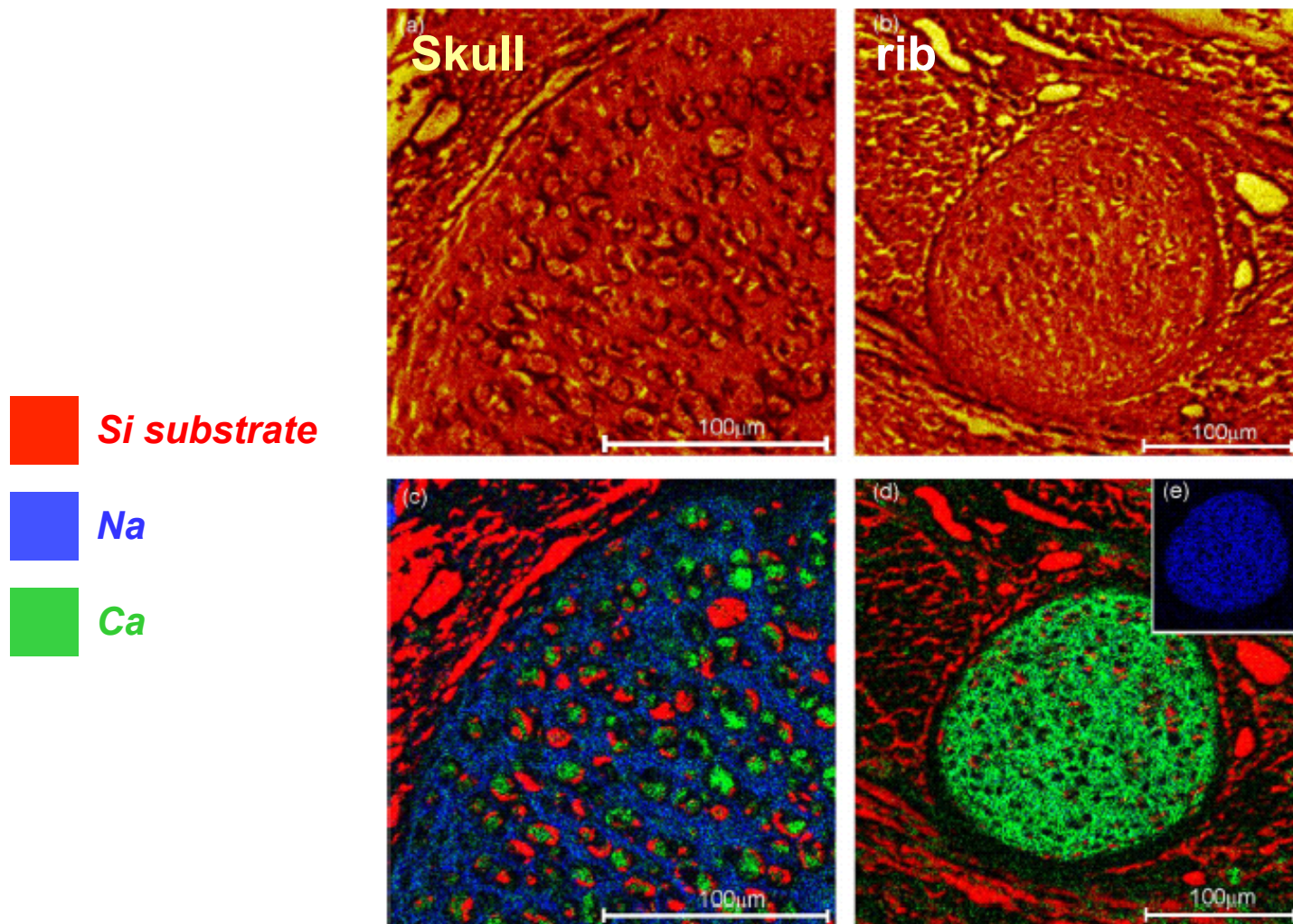
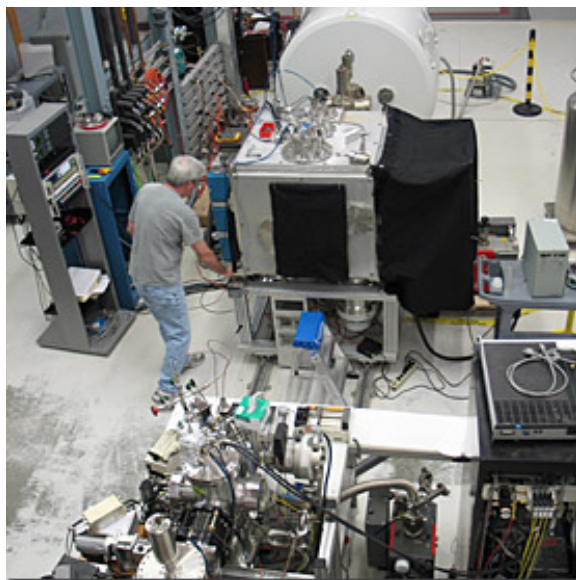


Fig. 5. (a) ToF-SIMS positive total ion image of a region of the skull. (b) ToF-SIMS positive total ion image of a rib. (c) Three-color overlay of the skull image; red color represents the Si substrate; green color represents the Na distribution; blue color represents the Ca distribution. Na and Ca are plotted on the same scale: (d) three-color overlay of the rib image, red color represents the Si substrate; green color represents the Na; blue color (too low to be visible) represents the Ca distribution. Na and Ca are plotted on the same scale: (e) Ca distribution with the signal intensity increased 5 \times .

How to choose a suitable MS?

- **Very low sample amount: go for MALDI.**
- **Require high mass accuracy: FTMS, TOF MS**
- **MS2 and high mass accuracy: TOF/TOF MS**
- **MS3 and above: ion trap (definitely FTICR if need high mass accuracy)**

Petrochemical Application of FT-ICRMS



A. G. Marshall, National High Magnetic Field Lab,
Florida State University, USA.

South American Petroleum Crude Oil
17,000+ Compositionally Distinct Components Resolved
by High Resolution 9.4 Tesla Electrospray FT-ICR MS

Negative Ions



Positive Ions

6,118 resolved
components

11,127 resolved
components

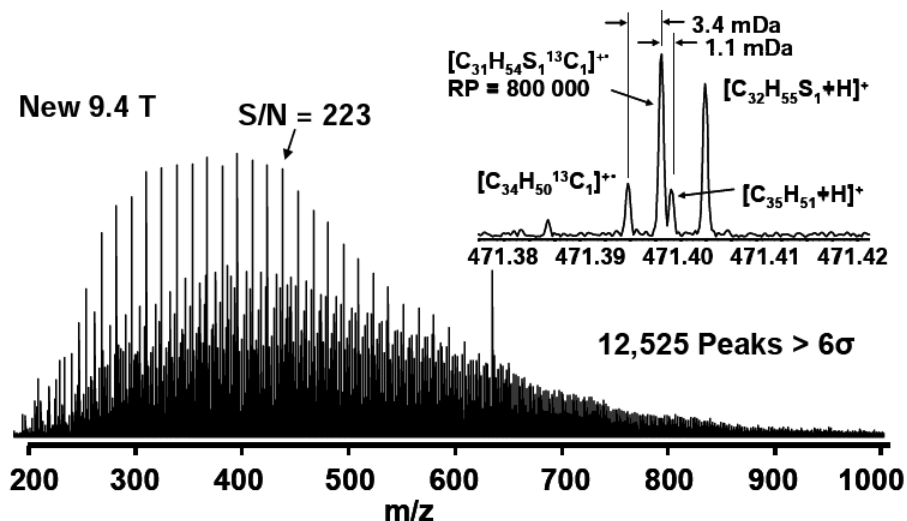
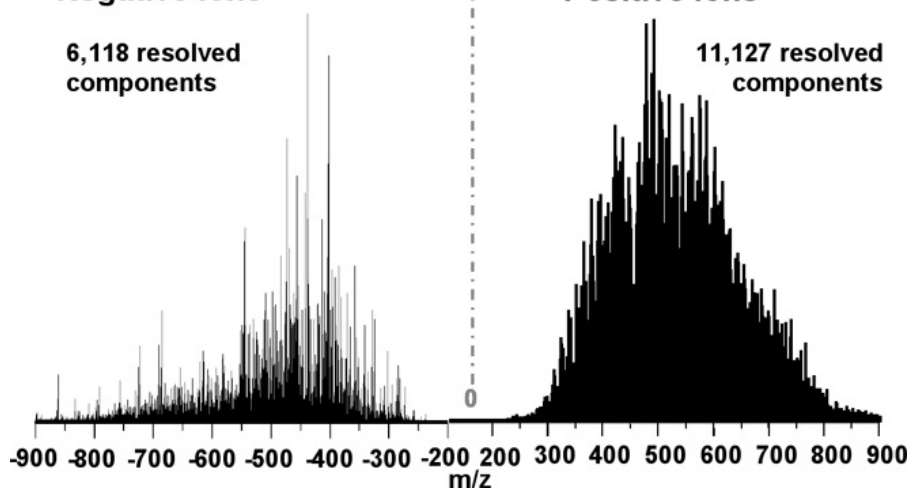


Figure 1. Positive-ion atmospheric pressure photoionization mass spectrum of a Middle Eastern light crude oil, acquired with the new NHMFL 9.4 T FT-ICR mass spectrometer. The inset illustrates resolution of the mass splits (1.1 mDa and 3.4 mDa) required for unequivocal identification of sulfur-containing components in petroleum heavy crude oils.

Ultra High Resolution FT-ICRMS



Evgenij *Nikolaev*



© American Society for Mass Spectrometry, 2011

J. Am. Soc. Mass Spectrom. (2011) 22:1125–1133
DOI: 10.1007/s13361-011-0125-9

RESEARCH ARTICLE

Initial Experimental Characterization of a New Ultra-High Resolution FTICR Cell with Dynamic Harmonization

Eugene N. Nikolaev,^{1,2,3} Ivan A. Boldin,^{1,2} Rolar

¹The Institute for Energy Problems of Chemical Physics, Russian Academy of Sciences, Moscow, Russia 119334

²Institute of Biochemical Physics, Russian Academy of Sciences, Moscow, Russia 119334

³The Institute of Biomedical Chemistry, Russian Academy of Medicine, Moscow, Russia 119134

⁴Bruker Daltonik GmbH, Bremen, Germany

Abstract

A new Fourier transform ion cyclotron resonance (FT-ICR) cell has been developed. The principles of formation of the effective electric potential and Nikolaev (Proceedings of the 58th ASMS Conference, 2011) is co-operation is based on the concept of electric potential averaging over the entire volume of a cylindrical Penning trap to be equivalent to the hyperbolic electrodes in an ideal Penning trap. The new cell are shaped for generating a quadratic dependence of the electric potential (along cyclotron motion orbit) electric potential at any radius. The electrodes together with the trapping segments form a cylindrical cell. In excitation mode this cell being a cylindrical cell of the same length. It is more effective than a Gabrielse et al.-type (Int J Mass Spectrom 1989) cylindrical cell with four compensation sections. Twenty millions of reserpine (m/z 609) and more than twenty millions of molecular ions (m/z 1357) has been obtained in a 7 T

Key words: FT ICR MS, Penning trap, Dynamic harmonization



Resolving Power: $> 10^7$

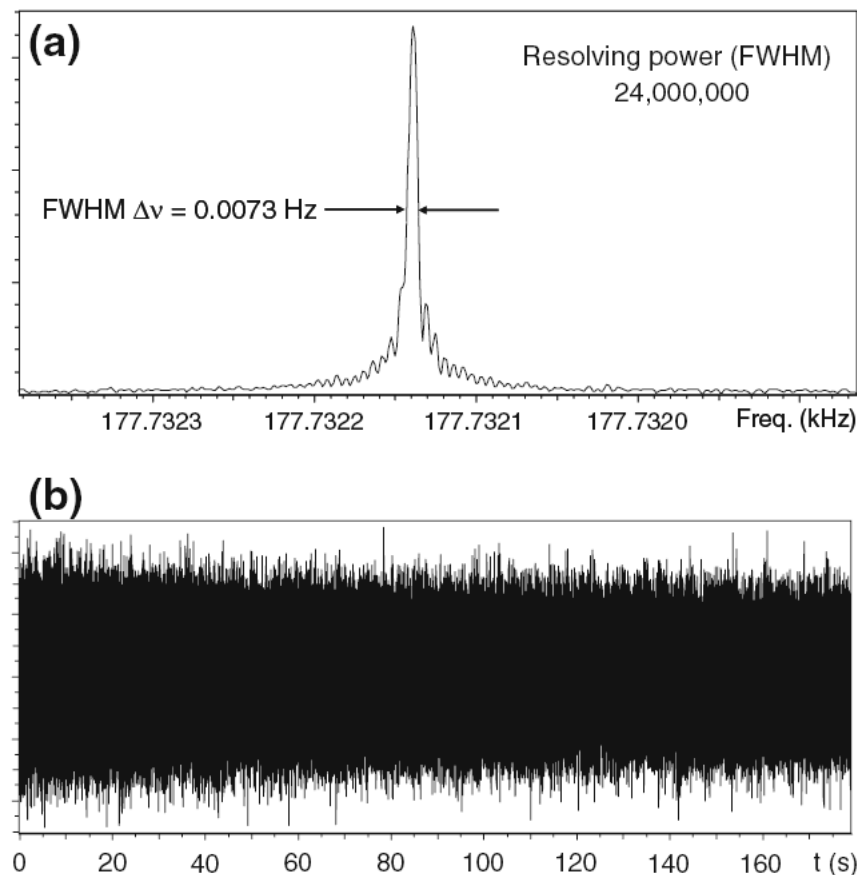


Figure 2. (a) Frequency spectrum (not calibrated) of the mono isotopic peak of singly charged, protonated reserpine (m/z 609.28066) with a resolving power of 24,000,000, resulting from magnitude FFT calculation without apodization. (b) Time domain spectrum of reserpine, detected over 3 min

Important Topics of MS Development

- **New ionization method for carbohydrates;**
- **New fragmentation methods, i.e. site-specific cleavage for oligosaccharides.**
- **Reliable method for quantitative measurements;**
- **Ion detector for high mass molecules;**
- **Miniaturization;**

Useful References

A. Books:

Mass spectrometry: Principles and applications

Edmond de Hoffmann, Vincent Stroobant, JohnWiley & Sons, LTD.,
Chichester, 2001.

Mass Spectrometry, A Test Book

Jurgen H. Gross, Springer, 2011.

B. Journal literatures:

Mike Tyers and Matthias Mann, Nature **422**, 193 (2003)

Bogdan Bogdanov and Richard D. Smith, Mass Spec. Rev. 24, 168(2005).

Gary L. Glish and Richard W. Vachet, Nature Rev. 2, 140 (2003).